

**3 - Organic Chemistry****UNIT TEST • PRACTICE**

- You may use a pencil, eraser, and scientific calculator to complete the test. No other resources are allowed.
- You have fifty (50) minutes to complete this test, unless other arrangements have been made.
- Please transfer your answers for Matching and Multiple Choice questions onto the Answer Document. Work for the Matching and Multiple Choice questions will not be graded. However, sufficient and appropriate work must be shown for the Free Response questions in order to receive full credit.

**SECTION 1: MATCHING.** Use the following answer choices that best match each description. Answer choices may be used once, more than once, or not at all. (1 point each)

**Questions 1-6:**

- (A) Alkanes only
- (B) Alkenes only
- (C) Alkynes only
- (D) Both alkenes and alkynes

1. Have a molecular formula  $C_2H_6$ .

2. Have *cis* - *trans* - isomers.

3. Can rotate around C-C bond.

4. Have a triple bond in the main carbon chain.

5. Have an H-C-C bond angle of  $120^\circ$ .

6.  $C_2H_6$  is an example.

7. Are considered "saturated" hydrocarbons.

**Questions 7-12:**

- (A) Ketone
- (B) Amine
- (C) Carboxylic Acid
- (D) Ether

8. Has a carbon bonded to a nitrogen.

9. Has a sour taste.

10. Acetone, or nail polish remover, is an example.

11. Has a carbon that is bonded to oxygen only with a double bond.

12. Has an oxygen single bonded to two carbons.

13. Vinegar is an example.

**SECTION 2: MULTIPLE CHOICE.** Select the best answer choice for each question. (2 points each)

14. Which of the following elements has a bonding capacity of 2?  
(A) Oxygen  
(B) Hydrogen  
(C) Carbon  
(D) Chlorine

15. Structural formulas differ from molecular formulas in that they show  
(A) The arrangement of atoms.  
(B) The bond angles in the molecule.  
(C) The elements that make up the compound.  
(D) The number of each element present.

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## **Table of Contents Chemistry Unit Test Practice**

1. Understanding the eBook Chemistry Unit Test Practice
  - The Rise of Digital Reading Chemistry Unit Test Practice
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit Test Practice
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Unit Test Practice
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit Test Practice
  - Personalized Recommendations
  - Chemistry Unit Test Practice User Reviews and Ratings
  - Chemistry Unit Test Practice and Bestseller Lists

5. Accessing Chemistry Unit Test Practice Free and Paid eBooks
  - Chemistry Unit Test Practice Public Domain eBooks
  - Chemistry Unit Test Practice eBook Subscription Services
  - Chemistry Unit Test Practice Budget-Friendly Options
6. Navigating Chemistry Unit Test Practice eBook Formats
  - ePUB, PDF, MOBI, and More
  - Chemistry Unit Test Practice Compatibility with Devices
  - Chemistry Unit Test Practice Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Unit Test Practice
  - Highlighting and Note-Taking Chemistry Unit Test Practice
  - Interactive Elements Chemistry Unit Test Practice
8. Staying Engaged with Chemistry Unit Test Practice
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Unit Test Practice
9. Balancing eBooks and Physical Books Chemistry Unit Test Practice
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Unit Test Practice
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Unit Test Practice
  - Setting Reading Goals Chemistry Unit Test Practice
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Unit Test Practice
  - Fact-Checking eBook Content of Chemistry Unit Test Practice
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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