

Multiple-Choice

Select the alternative that completes the statement or answers the question.

- _____ 1. A redox reaction is spontaneous if the standard reduction potential of the reductant is _____ the standard reduction potential of the oxidant.
- a. lower
b. the same
c. higher
d. unrelated
- _____ 2. Which of the following represents a polyatomic ion that carries a negative charge?
- a. SO_4^{2-}
b. NH_4^+
c. NO_3^-
d. H_2O
- _____ 3. Which species contains carbon with a positive oxidation state?
- a. CO
b. CO_2
c. C_2H_6
d. Na_2CO_3
- _____ 4. Which of the following is a weak electrolyte?
- a. NaOH
b. HCl
c. CH_3COOH
d. NaCl
- _____ 5. Which is a product of the combustion of a hydrocarbon? (Combustion refers to the reaction of a substance with O_2 .)
- a. water vapor
b. carbon dioxide
c. carbon monoxide
d. hydrogen gas
- _____ 6. Which molecule has a net dipole moment? (Indicate the direction of the net dipole moment.)
- a. CH_4 , south
b. CH_2Cl_2 , north
c. CH_2F_2 , north
d. CH_2Br_2 , south
- _____ 7. Which is the conjugate base of H_2PO_4^- in the reaction below?
- a. HPO_4^{2-}
b. H_2O
c. H_2PO_4^-
d. H_3PO_4
- _____ 8. Consider a solution of a substance (M_2X_3) at a concentration of 0.1 mol/L. Which of the following is the correct expression for the solubility product constant?
- a. $K_{sp} = 4x^3$
b. $K_{sp} = 27x^3$
c. $K_{sp} = 4x^4$
d. $K_{sp} = 27x^4$

Chem Semester Study Guide Multiple Choice

Jo Ann Robinson, Jacqueline E. Taylor



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