

## 7.3 Scientists traced the process of photosynthesis using isotopes

The leaves of plants that live in lakes and ponds are often covered with bubbles of O<sub>2</sub> produced during photosynthesis (Figure 7.3). How did scientists find out where this O<sub>2</sub> comes from?

The overall process of photosynthesis has been known since the 1800s: In the presence of light, green plants convert carbon dioxide and water into sugar and oxygen. Consider the basic summary equation for photosynthesis:



Looking at the equation, you can understand why scientists hypothesized that in photosynthesis, carbon dioxide is first split (CO<sub>2</sub>  $\rightarrow$  C + O<sub>2</sub>) and O<sub>2</sub> released, and then water (H<sub>2</sub>O) is added to the carbon to produce sugar. In the 1930s, this idea was challenged by C. B. van Niel, who was working with photosynthesizing bacteria that produce sugar from CO<sub>2</sub> but do not release O<sub>2</sub> in the process. These bacteria obviously did not split CO<sub>2</sub> in their photosynthesis. He hypothesized that in plant photosynthesis, it is H<sub>2</sub>O that is split, with the hydrogen becoming incorporated into sugar and the O<sub>2</sub> released as gas.

It was almost 20 years before van Niel's hypothesis was confirmed. Using a heavy isotope of oxygen, O-18, scientists were able to follow the fate of oxygen atoms during photosynthesis. (See Module 2.3 to review isotopes.) O-18 has two more neutrons in the nucleus of its atom than the more common isotope O-16, and this slight difference in mass can be experimentally measured. To determine the source of the O<sub>2</sub> released by photosynthesis, researchers produced CO<sub>2</sub> and H<sub>2</sub>O containing O-18, thus "labeling" the two reactant molecules. Experimental plants were provided with

one or the other labeled reactant. The results were that plants produced O<sub>2</sub> containing O-18 only when supplied with labeled H<sub>2</sub>O, never when provided with labeled CO<sub>2</sub>. These experiments showed that the O<sub>2</sub> released during photosynthesis comes from water and not from CO<sub>2</sub>.

The synthesis of sugar in photosynthesis involves numerous chemical reactions (as you will see in Module 7.10). Working out the details of these reactions also involved the use of isotopes, in this case, radioactive isotopes. In the mid-1940s, American biochemist Melvin Calvin and his colleagues began using radioactive C-14 to trace the sequence of the intermediate molecules formed in the cyclic pathway that produces sugar from CO<sub>2</sub>. They worked for 10 years to elucidate this cycle, which is now called the Calvin cycle. Calvin received the Nobel Prize in 1961 for this work.



▲ Figure 7.3 Oxygen bubbles on the leaves of an aquatic plant

Photosynthesis produces billions of tons of carbohydrate a year. Where does most of the mass of this huge amount of organic matter come from?

Additional text over question marks: "metabolizes glucose to release energy and uses oxygen atoms, which are in CO<sub>2</sub> when humans breathe out."

## 7.4 Photosynthesis is a redox process, as is cellular respiration

Let's compare the processes of photosynthesis and cellular respiration. Indeed, they appear to be opposite reactions: the reactants in one are the products of the other and vice versa. Both involve oxidation-reduction reactions (see

this energy boost). Figure 7.4 illustrates these energy-requiring redox reactions as CO<sub>2</sub> becomes reduced to sugar and water molecules are oxidized to O<sub>2</sub>.

# Chapter 7 Photosynthesis Study Guide

**Arshad Iqbal**

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