



Chemical Reactions

Section 1: Reactions and Equations

Chemical reactions are the rearrangement of atoms to produce different substances. There are six different types of chemical reactions:

- _____ 1. combination reactions
- _____ 2. decomposition reactions (one reactant, many products)
- _____ 3. precipitation reactions
- _____ 4. double displacement reactions (two reactants, two products)
- _____ 5. acid-base reactions
- _____ 6. oxidation-reduction reactions
- _____ 7. reduction reactions (oxidation and reduction reactions)
- _____ 8. Redox reactions (oxidation and reduction reactions)

Describe the following reaction using one of the above. Include the reaction type, reactants, and products.

Reactants	Products	Reactants	Products
Hydrogen gas + Chlorine gas	Hydrochloric acid	Iron + Oxygen	Iron(III) oxide

Draw four lines from the above column to the column below to complete the following reaction equations. Use one line for each reaction type.

1. _____ + _____ → _____ + _____ (acid-base reaction)

2. _____ + _____ → _____ + _____ (double displacement reaction)

3. _____ + _____ → _____ + _____ (reduction reaction)

4. _____ + _____ → _____ + _____ (oxidation reaction)

5. _____ + _____ → _____ + _____ (reduction reaction)

6. _____ + _____ → _____ + _____ (oxidation reaction)



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Robert J. Ouellette, J. David Rawn

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Chemistry:

Matter & Change, Study Guide For Content Mastery, Student Edition McGraw Hill,2001-03-30 Study Guide and Reinforcement Worksheets allow for differentiated instruction through a wide range of question formats There are worksheets and study tools for each section of the text that help teachers track students progress toward understanding concepts Guided Reading Activities help students identify and comprehend the important information in each chapter

Chemistry ,2001-04-01 **Handbook of Reading Research, Volume III** Michael L. Kamil,Peter B. Mosenthal,P. David Pearson,Rebecca Barr,2016-11-18 In Volume III as in Volumes I and II the classic topics of reading are included from vocabulary and comprehension to reading instruction in the classroom and in addition each contributor was asked to include a brief history that chronicles the legacies within each of the volume's many topics However on the whole Volume III is not about tradition Rather it explores the verges of reading research between the time Volume II was published in 1991 and the research conducted after this date The editors identified two broad themes as representing the myriad of verges that have emerged since Volumes I and II were published 1 broadening the definition of reading and 2 broadening the reading research program The particulars of these new themes and topics are addressed

Prentice Hall Chemistry ,2000 Organic

Chemistry Francis A. Carey,2000 A Market Leading Traditional Approach to Organic Chemistry Throughout all seven editions Organic Chemistry has been designed to meet the needs of the mainstream two semester undergraduate organic chemistry course This best selling text gives students a solid understanding of organic chemistry by stressing how fundamental reaction mechanisms function and reactions occur With the addition of handwritten solutions new cutting edge molecular illustrations updated spectroscopy coverage seamless integration of molecular modeling exercises and state of the art multimedia tools the 7th edition of Organic Chemistry clearly offers the most up to date approach to the study of organic chemistry

Resources in Education ,2000-04 **Saunders Interactive General Chemistry CD-ROM** Mary L Kotz,John C. Kotz,William J. Vining,Justin T. Fermann,1998-11 **Contemporary Educational Management in East Africa** Stuart Busby,Chrispina Lekule,Christina Raphael,2026-02-09 This rigorous and insightful volume provides comprehensive analysis and research findings on educational management challenges in East Africa homing in on the interrelated areas of inclusion gender equity leadership and digital integration in schools and higher education institutions to foster a more equitable and

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Teacher's Guide and Key for Broadening Worlds of Science Willard J. Jacobson,1959 **The Science Teachers Bulletin** ,1936 **Understanding Organic Chemistry** Margot K. Schumm,1987 **Chemistry** Steven S.

Zumdahl,Susan A. Zumdahl,2000 Zumdahl s conceptual approach to problem solving early and thorough treatment of reactions and integration of descriptive chemistry and chemical principles throughout are key elements to the success of this bestseller By encouraging students to understand qualitative concepts before considering quantitative problems Chemistry helps students build a solid understanding of topics and avoid common misconceptions The text also includes new Chemical Impact boxes that present everyday applications of chemistry plus over 200 redesigned illustrations with greater clarity and emphasis on molecular level illustrations

Organic Chemistry Study Guide Robert J. Ouellette,J. David

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