

CHAPTER 4 PRACTICE TEST: SOLUTION STOICHIOMETRY

1) Which of the following are weak electrolytes? HCl, $\text{HC}_2\text{H}_3\text{O}_2$, NH_3 , KCl
 A) HCl, KCl
 B) HCl, $\text{HC}_2\text{H}_3\text{O}_2$, NH_3 , KCl
 C) $\text{HC}_2\text{H}_3\text{O}_2$, KCl
 D) $\text{HC}_2\text{H}_3\text{O}_2$, NH_3
 E) HCl, $\text{HC}_2\text{H}_3\text{O}_2$, KCl

Answer: D Diff: 2

2) What are the spectator ions in the reaction between KOH (aq) and HNO_3 (aq)?

A) K^+ and H^+ C) K^+ and NO_3^- E) OH^- only
 B) H^+ and OH^- D) H^+ and NO_3^-

Answer: C Diff: 2

3) The balanced net ionic equation for precipitation of CaCO_3 when aqueous solutions of Na_2CO_3 and CaCl_2 are mixed is _____.

A) 2Na^+ (aq) + CO_3^{2-} (aq) \rightarrow Na_2CO_3 (aq)
 B) 2Na^+ (aq) + 2Cl^- (aq) \rightarrow 2NaCl (aq)
 C) Na^+ (aq) + Cl^- (aq) \rightarrow NaCl (aq)
 D) Ca^{2+} (aq) + CO_3^{2-} (aq) \rightarrow CaCO_3 (s)

Answer: D Diff: 3

4) Which of the following are strong acids? HI, HNO_3 , HF, HBr

A) HF, HBr C) HI, HF, HBr E) HI, HNO_3 , HBr
 B) HI, HNO_3 , HF, HBr D) HNO_3 , HF, HBr

Answer: E Diff: 2

5) A neutralization reaction between an acid and a metal hydroxide produces _____.

A) water and a salt C) oxygen gas E) ammonia
 B) hydrogen gas D) sodium hydroxide

Answer: A Diff: 2

6) Which of these metals is the least easily oxidized? Na, Au, Fe, Ca, Ag

A) Na B) Au C) Fe D) Ca E) Ag

Answer: B Diff: 3

7) Based on the equations below, which metal is the most active?



A) Ni B) Ag C) Cu D) Pb E) N

Answer: A Diff: 3

8) What is the concentration (M) of KCl in a solution made by mixing 25.0 mL of 0.100 M KCl with 50.0 mL of 0.100 M KCl?

A) 0.100 B) 0.0500 C) 0.0333 D) 0.0250 E) 125

Answer: A Diff: 3

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