

Name: \_\_\_\_\_

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## Student Exploration: Covalent Bonds

**Vocabulary:** covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? **They can share the markers.**

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

**They can give the ones they need to each other.**

### Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the **Covalent Bonds** Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** (▶) to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? \_\_\_\_\_ 7

2. Click **Pause** (⏸). Drag an electron from the left atom to the right atom. Click **Play**. What happens? **They share an electron.**

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**. What happens now? **They share another one.**

# Covalent Bond Gizmo Answer Key

**Lei Huang**

**Covalent Bond Gizmo Answer Key:**

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## **Table of Contents Covalent Bond Gizmo Answer Key**

1. Understanding the eBook Covalent Bond Gizmo Answer Key
  - The Rise of Digital Reading Covalent Bond Gizmo Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Covalent Bond Gizmo Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Covalent Bond Gizmo Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Covalent Bond Gizmo Answer Key
  - Personalized Recommendations
  - Covalent Bond Gizmo Answer Key User Reviews and Ratings
  - Covalent Bond Gizmo Answer Key and Bestseller Lists
5. Accessing Covalent Bond Gizmo Answer Key Free and Paid eBooks

- Covalent Bond Gizmo Answer Key Public Domain eBooks
- Covalent Bond Gizmo Answer Key eBook Subscription Services
- Covalent Bond Gizmo Answer Key Budget-Friendly Options

6. Navigating Covalent Bond Gizmo Answer Key eBook Formats

- ePUB, PDF, MOBI, and More
- Covalent Bond Gizmo Answer Key Compatibility with Devices
- Covalent Bond Gizmo Answer Key Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Covalent Bond Gizmo Answer Key
- Highlighting and Note-Taking Covalent Bond Gizmo Answer Key
- Interactive Elements Covalent Bond Gizmo Answer Key

8. Staying Engaged with Covalent Bond Gizmo Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Covalent Bond Gizmo Answer Key

9. Balancing eBooks and Physical Books Covalent Bond Gizmo Answer Key

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Covalent Bond Gizmo Answer Key

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Covalent Bond Gizmo Answer Key

- Setting Reading Goals Covalent Bond Gizmo Answer Key
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Covalent Bond Gizmo Answer Key

- Fact-Checking eBook Content of Covalent Bond Gizmo Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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