

# Covalent bonding and Lewis electron diagrams – Notes

## Learning Intention

We will be learning how to represent the compounds formed between two non-metals using Lewis electron diagrams

## Success Criteria

I can:

- ☐ Relate electrons being shared to covalent bonds
- ☐ Draw Lewis electron dot structures for covalent compounds
- ☐ Correctly use naming rules for covalent compounds



## Warm Up Question

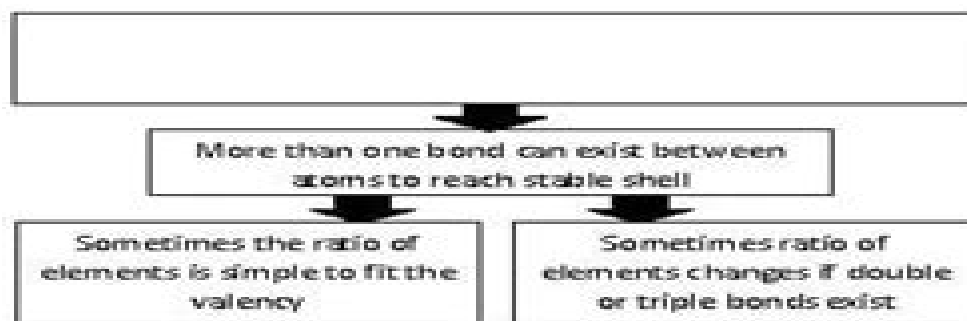
Consider the two elements phosphorus and fluorine

1. Draw separate Lewis electron dot diagrams for each.
2. Based on your diagrams, describe how each of these elements would reach a stable valence shell fastest.

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## Covalent



## Prediction of compound structure question

Consider the two elements phosphorus and fluorine

1. What is the simplest ratio these elements would occur in?
2. Draw a Lewis diagram to predict how these elements might be combined in a compound.

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# Covalent Bonding Guided Answers

**Leonard W. Fine**



## Covalent Bonding Guided Answers:

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