

## Chapter 19

1. a) Contains as many hydrogen atoms as possible for the given number of carbons. Only carbon-carbon single bonds.

b) (i)  $C_{11}H_{24}$   
 (ii) liquid  
 (iii)  $C_{11}H_{24}(l) + 17O_2(g) \longrightarrow 11CO_2(g) + 12H_2O(l)$

2. a) a carbon-carbon double bond

b)

$$\begin{array}{c} \text{H} & \text{H} \\ | & | \\ \text{H}-\text{C} & -\text{C}=\text{C} \\ | & | \\ \text{H} & \text{H} \end{array}$$

c)

$$\begin{array}{c} \text{H} & \text{H} \\ | & | \\ \text{H}-\text{C} & -\text{C}=\text{C} \\ | & | \\ \text{H} & \text{H} \end{array} + Br_2 \longrightarrow \begin{array}{c} \text{H} & Br & \text{H} \\ | & | & | \\ \text{H}-\text{C} & -\text{C} & -\text{C}-Br \\ | & | & | \\ \text{H} & \text{H} & \text{H} \end{array}$$

(Home schooling parents: It doesn't matter what position you draw the bromine in on the right-hand carbon - IN, S or E (with hydrogens in the other two positions). All that matters is that one bromine becomes attached to each of the carbons either side of the original double bond.)

d) (i)  $CH_4(g) + Br_2(g) \longrightarrow CH_3Br(g) + HBr(g)$   
 (ii) In this (substitution) reaction, one of the hydrogens has been replaced by a bromine during the reaction. In the previous (addition) reaction, nothing was lost when the two molecules combined together.

3. a) Ethene is passed with steam at  $300^{\circ}\text{C}$  and 60 - 70 atmospheres over a phosphoric acid catalyst.  
 b) e.g. two off: continuous flow process, therefore faster; rapid reaction; purer product.  
 c) uses renewable resources; gentle conditions.

4. Dissolve the sugar in water in a conical flask (or similar) and add yeast. You now need some means of keeping the air out while allowing carbon dioxide to escape. You could use a commercial home-winemaking airlock of some sort, but a plug of cotton wool works just as well. Leave the mixture in the warm for a few days until bubbling stops.



Allow the yeast to settle, and pour off the solution so that it can be fractionally distilled, collecting the fraction which boils at  $78^{\circ}\text{C}$ . (See the diagram for the fractional distillation of an alcohol/water mixture in Chapter 11 of the book.)

continued . . .

# Catalyst Laboratory Manual Answers

**Michael Seilmaier**

## **Catalyst Laboratory Manual Answers:**

*Basic Concepts of Chemistry* Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance.

**Lab Manual for General, Organic, and Biochemistry** Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging. Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives. *Essentials of General Organic and Biochemistry* captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course. Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession. To browse a sample chapter view sample ChemCasts and more visit [www.whfreeman.com/gob](http://www.whfreeman.com/gob)

*Study Guide/Selected Solutions Manual* Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book.

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one term courses covering the whole spectrum of any part of it The experiments have been used by students and have been frequently revised following student polls regarding clarity and interest and suggestions from instructors The questions on the Report and Observation Sheets have again been adjusted in the light of student comments and more room for answers has been provided on many Report Sheets Prentice Hall Chemistry,2000 **Instructor's Manual** Harold Hart,1983

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