

# 10.1 Periodic Trends



## 10.1 PERIODIC TRENDS FROM CONCEPT MASTERY

### Section 10.1 (continued)

In your notebook, read aloud the student periodicals table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	21
_____	22
_____	23
_____	24

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

#### Column A

- \_\_\_\_\_ 19. It increases in the periodic table.
- \_\_\_\_\_ 20. It decreases in the periodic table.
- \_\_\_\_\_ 21. Group 1 elements.
- \_\_\_\_\_ 22. Elements that are shiny and conduct electricity.
- \_\_\_\_\_ 23. Group 17 elements.

#### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word as shown in smaller letters.

- \_\_\_\_\_ 24. There are two representative elements of elements.
- \_\_\_\_\_ 25. Most elements are metals in the periodic table.
- \_\_\_\_\_ 26. Group 1 elements except for hydrogen are metals in the periodic table.
- \_\_\_\_\_ 27. Group 17 elements are the active earth metals.
- \_\_\_\_\_ 28. Group 1 elements are highly reactive elements in the periodic table.
- \_\_\_\_\_ 29. Group 17 elements are very reactive elements found in the periodic table.
- \_\_\_\_\_ 30. Elements have properties of both metals and nonmetals.

# Content Mastery Study Guide Key

**Philip Tate, Rodney R. Seeley, James R.  
Kennedy, Trent D. Stephens**

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## **Table of Contents Content Mastery Study Guide Key**

1. Understanding the eBook Content Mastery Study Guide Key
  - The Rise of Digital Reading Content Mastery Study Guide Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Content Mastery Study Guide Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Content Mastery Study Guide Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Content Mastery Study Guide Key
  - Personalized Recommendations
  - Content Mastery Study Guide Key User Reviews and Ratings
  - Content Mastery Study Guide Key and Bestseller Lists
5. Accessing Content Mastery Study Guide Key Free and Paid eBooks
  - Content Mastery Study Guide Key Public Domain eBooks
  - Content Mastery Study Guide Key eBook Subscription Services
  - Content Mastery Study Guide Key Budget-Friendly Options



6. Navigating Content Mastery Study Guide Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Content Mastery Study Guide Key Compatibility with Devices
  - Content Mastery Study Guide Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Content Mastery Study Guide Key
  - Highlighting and Note-Taking Content Mastery Study Guide Key
  - Interactive Elements Content Mastery Study Guide Key
8. Staying Engaged with Content Mastery Study Guide Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Content Mastery Study Guide Key
9. Balancing eBooks and Physical Books Content Mastery Study Guide Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Content Mastery Study Guide Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Content Mastery Study Guide Key
  - Setting Reading Goals Content Mastery Study Guide Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Content Mastery Study Guide Key
  - Fact-Checking eBook Content of Content Mastery Study Guide Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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