

Chapter 1: Introduction

1-13. The value $g = 9.81 \text{ m/s}^2$ is specific to the force of gravity on the surface of the earth. The universal formula for the force of gravitational attraction is:

$$F = G \frac{m_1 m_2}{r^2}$$

Where m_1 and m_2 are the masses of the two objects, r is the distance between the centers of the two objects, and G is the universal gravitation constant, $G = 6.674 \times 10^{-11} \text{ N(m/kg)}^2$.

- Research the diameters and masses of the Earth and Jupiter.
- Demonstrate that $F = m(9.81 \text{ m/s}^2)$ is a valid relationship on the surface of the earth.
- Determine the force of gravity acting on a 1000 kg satellite that is 2000 miles above the surface of the Earth.
- One of the authors of this book has a mass of 200 lb_m. If he was on the surface of Jupiter, what gravitational force in lb_f would be acting on him?

Solution:

- A. Measurements obtained from different sources will vary slightly.

$$D_{\text{Earth}} = 12,742 \text{ km}$$

$$D_{\text{Jupiter}} = 142,000 \text{ km}$$

$$\text{Mass}_{\text{Earth}} = 5.97 \times 10^{24} \text{ kg}$$

$$\text{Mass}_{\text{Jupiter}} = 1.90 \times 10^{27} \text{ kg}$$

- B. $\text{Mass}_{\text{Earth}} = 5.97 \times 10^{24} \text{ kg}$ $\text{Radius}_{\text{Earth}} = 6.371 \times 10^6 \text{ meters}$

$$F = G \frac{m_1 m_2}{r^2} \rightarrow F = m_{\text{object}} \left(6.674 \times 10^{-11} \frac{\text{N} \cdot \text{m}^2}{\text{kg}^2} \right) \frac{5.97 \times 10^{24} \text{ kg}}{(6.37 \times 10^6 \text{ m})^2} \left(\frac{\text{N} \cdot \text{s}^2}{\text{kg} \cdot \text{m}} \right)$$

$$\therefore F = m \left(9.81 \frac{\text{m}}{\text{s}^2} \right)$$

- C. Using the universal gravitational attraction formula:

$$F = G \frac{m_1 m_2}{r^2}$$

$$F = \left(6.674 \times 10^{-11} \frac{\text{N} \cdot \text{m}^2}{\text{kg}^2} \right) \frac{(1000 \text{ kg})(5.97 \times 10^{24} \text{ kg})}{(6.371 \times 10^6 \text{ m} + 3200 \text{ km})^2}$$

$$F = 4350 \text{ N}$$

- D. $\text{Radius}_{\text{Jupiter}} = 71000000 \text{ m}$ $\text{Mass}_{\text{Jupiter}} = 1.898 \times 10^{27} \text{ kg}$

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RJ Alexander



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Solutions Manual - Chemistry Charles E. Mortimer, 1986 *Whitaker's Book List*, 1991 *Whitaker's Cumulative Book List*, 1986 *Experimental Methods in Wastewater Treatment* Mark C. M. van Loosdrecht, Per Halkjaer Nielsen, C. M. Lopez-Vazquez, Damir Brdjanovic, 2016-05-15 Over the past twenty years the knowledge and understanding of wastewater treatment has advanced extensively and moved away from empirically based approaches to a fundamentally based first principles approach embracing chemistry microbiology and physical and bioprocess engineering often involving experimental laboratory work and techniques Many of these experimental methods and techniques have matured to the degree that they have been accepted as reliable tools in wastewater treatment research and practice For sector professionals especially a new generation of young scientists and engineers entering the wastewater treatment profession the quantity complexity and diversity of these new developments can be overwhelming particularly in developing countries where access to advanced level laboratory courses in wastewater treatment is not readily available In addition information on innovative experimental methods is scattered across scientific literature and only partially available in the form of textbooks or guidelines This book seeks to address these deficiencies It assembles and integrates the innovative experimental methods developed by research groups and practitioners around the world *Experimental Methods in Wastewater Treatment* forms part of the internet based curriculum in wastewater treatment at UNESCO IHE and as such may also be used together with video records of experimental methods performed and narrated by the authors including guidelines on what to do and what not to do The book is written for undergraduate and postgraduate students researchers laboratory staff plant operators consultants and other sector professionals **Pesticides Abstracts**, 1981 **The Bookseller, Newsdealer and Stationer**, 1899 **Bookseller and the Stationery Trades' Journal**, 1982 *The Chemical News and Journal of Physical Science*, 1904 **Solutions of the Cambridge Senate-house Problems and Riders for the Year 1875** Sir George Greenhill, 1876 *Solutions of Weekly Problem Papers* John James Milne, 1885 *The Bookseller*, 1981 Vols for 1871 76 1913 14 include an extra number The Christmas bookseller separately paged and not included in the consecutive numbering of the regular series *Scientific and Technical Aerospace Reports*, 1976 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database **Chemical Engineering Education**, 2001 *Health Aspects of Pesticides Abstract Bulletin*, 1981 **Paperbacks in Print**, 1979 **Engineering Digest**, 1988 **Weekly problem papers, with notes. [With] Solutions** John James Milne, 1885 **The Publishers' Trade List Annual**, 1985 *Engineering and Chemical Thermodynamics* Milo D. Koretsky, 2012-12-17 Koretsky helps students understand and visualize thermodynamics through a qualitative discussion of the role of molecular interactions and a highly visual presentation of the material By showing how principles of thermodynamics relate to molecular concepts learned in prior courses Engineering and Chemical

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