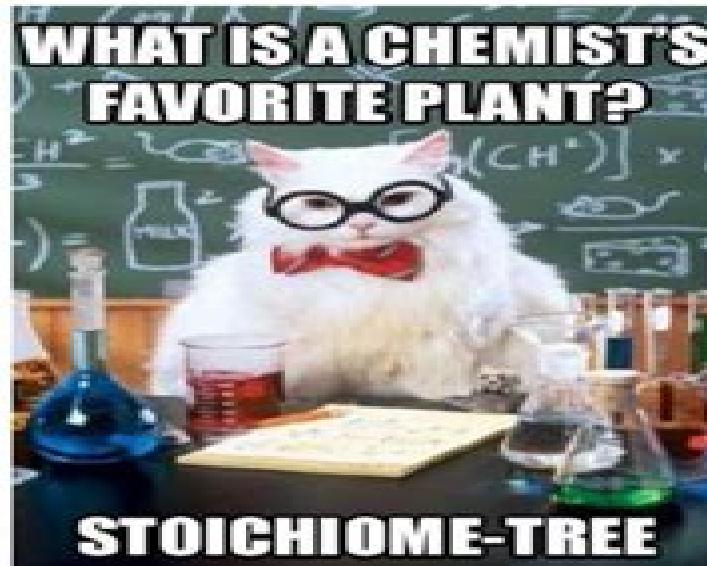


Chapter 11: Stoichiometry



Particle and Mole Relationships

- Chemical reactions stop when one of the reactants is used up
- **Stoichiometry**: the study of quantitative relationships between the amounts of reactants used and amounts of products formed by a chemical reaction
- The amount of each reactant present at the start of a chemical reaction determines how much product can form
- Based on the **law of conservation of mass**
 - Matter is neither created or destroyed in a chemical reaction

Chapter 11 Study Guide Stoichiometry Answer Key

Rose Arny

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- Chapter 2: Essential Elements of Chapter 11 Study Guide Stoichiometry Answer Key
- Chapter 3: Chapter 11 Study Guide Stoichiometry Answer Key in Everyday Life
- Chapter 4: Chapter 11 Study Guide Stoichiometry Answer Key in Specific Contexts
- Chapter 5: Conclusion

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