



~~1000000000~~  $109 / 17.679 \times 100 = 56.6\%$



5. a. TRUE  
b. FALSE  
c.  $\text{AgNO}_3 \rightarrow 169.91 \text{ g/mol}$   
d.  $6.02 \times 10^{23}$  particles = 1 mole of particles

# Chapter 11 Stoichiometry Study Guide Answers

**Donald W. Shive, Louise E. Shive**

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