

Chapter 25 Nuclear Chemistry Practice Test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- What does the 4 in ${}_{2}^{4}\text{He}$ represent?
 - the mass number
 - the atomic number
 - the number of protons
 - the number of neutrons
- What does the 101 in ${}_{101}^{258}\text{Md}$ represent?
 - the mass number
 - the atomic number
 - the nuclide number
 - the number of neutrons
- Balance the following equation: ${}_{38}^{226}\text{Rn} \rightarrow {}_{36}^{222}\text{Rn} + \underline{\hspace{2cm}}$
 - ${}_{2}^{4}\text{He}$
 - ${}_{0}^{1}\text{e}^{-}$
 - ${}_{1}^{1}\text{H}$
 - ${}_{-1}^{0}\text{e}^{+}$
- Balance the following equation: ${}_{93}^{239}\text{Np} \rightarrow \underline{\hspace{2cm}} + {}_{-1}^{0}\text{e}^{-}$
 - ${}_{90}^{239}\text{Th}$
 - ${}_{92}^{239}\text{U}$
 - ${}_{94}^{239}\text{Pu}$
 - ${}_{92}^{239}\text{Pu}$
- The spontaneous disintegration of a nucleus into a slightly lighter and more stable nucleus, accompanied by emission of particles, electromagnetic radiation, or both, is
 - nuclear fusion.
 - nuclear radiation.
 - radioactive decay.
 - nuclear fission.
- Which of the following forms of radiation has the greatest penetrating power?
 - alpha particles
 - beta particles
 - gamma rays
 - positrons
- Alpha particles are
 - electrons.
 - helium nuclei.
 - electromagnetic waves.
 - neutrons.
- Beta particles are
 - electrons.
 - helium nuclei.
 - electromagnetic waves.
 - neutrons.
- Which of the illustrations below represents a fission reaction?

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H. Stephen Stoker,Danny V. White

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