

## AP Biology Study Guide 1

### Water Properties:

Cohesion- Hydrogen bonding between **LIKE** molecules  
(ex: water with water)

Adhesion- Hydrogen bonding between **UNLIKE** molecules  
(ex: water with a leaf)

Capillary Action- the movement of a liquid through a narrow space as a result of the forces of cohesion, adhesion, and surface tension.  
(ex: Moving ground water from wet areas of soil to dry ones.)

Universal Solvent- Water is referred to as a universal solvent because it dissolves more substances than any other liquid.

Hydration Shell- the sphere of water molecules around each dissolved ion.

Moderation of Temperature- water moderates air temperature by absorbing heat from the air that is warmer and releasing stored heat to the air that is cooler.

Density- water is **LESS** dense as a **SOLID** than a **LIQUID**  
(this is because of water's hydrogen bonds)

Polarity- water is a polar molecule because of the unequal distribution of electrons.

High Specific Heat- water has an unusually **HIGH** specific heat, this is how water is able to maintain its homeostasis and keep the ocean habitable.

### Macromolecules:

#### Carbohydrates:

Elemental Makeup: Carbon, Hydrogen and Oxygen

Monomer: Monosaccharides (ex: Glucose & Fructose)

#### Polysaccharides:

Cellulose: Strengthens cell walls

Starch: Stores glucose for energy (plants)

Glycogen: Stores glucose for energy (animals)

Chitin: Strengthens exoskeletons and fungal cell walls

Key Function: Fuel, Carbon source to be converted to other molecules or into polymers

# Chapter 36 Ap Biology Answers For Guide

**G Orfield**

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**Social Workers' Desk Reference** Albert R. Roberts, Gilbert J. Greene, 2002 Following in the groundbreaking path of its predecessor the second edition of the Social Workers Desk Reference provides reliable and highly accessible information about effective services and treatment approaches across the full spectrum of social work practice

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## **Table of Contents Chapter 36 Ap Biology Answers For Guide**

1. Understanding the eBook Chapter 36 Ap Biology Answers For Guide
  - The Rise of Digital Reading Chapter 36 Ap Biology Answers For Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 36 Ap Biology Answers For Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 36 Ap Biology Answers For Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 36 Ap Biology Answers For Guide
  - Personalized Recommendations

- Chapter 36 Ap Biology Answers For Guide User Reviews and Ratings
- Chapter 36 Ap Biology Answers For Guide and Bestseller Lists

5. Accessing Chapter 36 Ap Biology Answers For Guide Free and Paid eBooks

- Chapter 36 Ap Biology Answers For Guide Public Domain eBooks
- Chapter 36 Ap Biology Answers For Guide eBook Subscription Services
- Chapter 36 Ap Biology Answers For Guide Budget-Friendly Options

6. Navigating Chapter 36 Ap Biology Answers For Guide eBook Formats

- ePUB, PDF, MOBI, and More
- Chapter 36 Ap Biology Answers For Guide Compatibility with Devices
- Chapter 36 Ap Biology Answers For Guide Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Chapter 36 Ap Biology Answers For Guide
- Highlighting and Note-Taking Chapter 36 Ap Biology Answers For Guide
- Interactive Elements Chapter 36 Ap Biology Answers For Guide

8. Staying Engaged with Chapter 36 Ap Biology Answers For Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chapter 36 Ap Biology Answers For Guide

9. Balancing eBooks and Physical Books Chapter 36 Ap Biology Answers For Guide

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Chapter 36 Ap Biology Answers For Guide

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Chapter 36 Ap Biology Answers For Guide

- Setting Reading Goals Chapter 36 Ap Biology Answers For Guide
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chapter 36 Ap Biology Answers For Guide

- Fact-Checking eBook Content of Chapter 36 Ap Biology Answers For Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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