

Concentration Worksheet – Answers

- 1) How many grams of beryllium chloride are needed to make 125 mL of a 0.050 *M* solution?
0.50 grams
- 2) How many grams of beryllium chloride would you need to add to 125 mL of water to make a 0.050 molal solution?
0.50 grams
- 3) The density of ethanol is 0.789 g/mL. How many grams of ethanol should be mixed with 225 mL of water to make a 4.5% (v/v) mixture?
7.99 grams
- 4) Explain how to make at least one liter of a 1.25 molal ammonium hydroxide solution.
Dissolve 43.8 grams (1.25 moles) of ammonium hydroxide in 1 L H₂O.
- 5) What is the molarity of a solution in which 0.45 grams of sodium nitrate are dissolved in 265 mL of solution.
0.020 *M*
- 6) What is the mole fraction of sulfuric acid in a solution made by adding 3.4 grams of sulfuric acid to 3,500 mL of water?
1.8 × 10⁻⁴
- 7) What will the volume of a 0.50 *M* solution be if it contains 25 grams of calcium hydroxide?
680 mL
- 8) How many grams of ammonia are present in 5.0 L of a 0.050 *M* solution?
4.3 grams

Chemistry Study Guide Solution Concentration Answers

Lauren Gardner

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Organic Chemistry Study Guide and Solutions Marc Loudon, Jim Parise, 2015-07-01 Parise and Loudon's Study Guide and Solutions Manual offers the following learning aids: Links that provide hints for study approaches to problem solving and additional explanations of challenging topics; Further Explorations that provide additional depth on key topics; Reaction summaries that delve into key mechanisms and stereochemistry; Solutions to all the textbook problems. Rather than providing just the answer, many of the solutions provide detailed explanations of how the problem should be approached.

Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl, Paul B. Kelter, 1993

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Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even-numbered end-of-chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled "Tying It All Together with a Laboratory Application".

Introduction to Chemistry, Study Guide T. R. Dickson, 1999-07-07 Teaches chemistry by offering a dynamic, provocative, and relevant view of the topic and its importance to society and our daily lives. Three themes are stressed throughout the text: developing chemical thinking, chemical vision, and refining problem-solving skills. Many chapters in this edition have been rewritten and rearranged to vitalize the topics and to include interesting examples, analogies, and images.

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