

Properties of Water



Water has the unique ability to dissolve more substances than any other liquid because of its strong, polar nature.

Water is a polar molecule, which means it has a partial positive charge on one side and a partial negative charge on the other.



Water has a relatively high density compared to most other substances.



Water has a relatively high melting and boiling point.



Water has a relatively high surface tension and viscosity.

Water has a relatively high heat capacity and heat of vaporization.

Water has a relatively high density of vapor.

2.2 - WATER

2.2.1 - Chemical Characteristics of Water

Water is a polar molecule because it contains two hydrogen atoms and one oxygen atom. It has a partial positive charge on each hydrogen atom and a partial negative charge on the oxygen atom.



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Sabine Zange

Chemistry Uniquely Water Study Guide Answers:

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh,2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

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