

Learning Objectives

Problem 1:1 Objectives

- ✓ Understand the concept of water's unique physical properties—density, specific heat, and surface tension.
- ✓ Explain how the structure of water leads to its unique properties.
- ✓ Understand the difference between polar and nonpolar molecules and how this affects their solubility in water.
- ✓ Explain the difference between hydrophilic and hydrophobic molecules.
- ✓ Understand the concept of water's high specific heat and how this affects its role in temperature regulation.
- ✓ Explain the concept of water's high surface tension and how this affects its role in capillary action.
- ✓ Explain the concept of water's high boiling point and how this affects its role in the water cycle.

2.1 – WATER

2.1.1 – Chemical Characteristics of Water

- Understand the chemical structure of water and how it leads to its unique properties. The water molecule is composed of two hydrogen atoms and one oxygen atom. The oxygen atom is more electronegative than the hydrogen atoms, leading to a polar covalent bond. This results in a partial negative charge on the oxygen atom and a partial positive charge on the hydrogen atoms.



Chemistry Uniquely Water Study Guide Answers

Sabine Zange



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Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application **Chemistry** John S. Phillips, Cheryl Wistrom, 2000 **Synthesis of**

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Unique Metal Ion Structures in Solid Matrices Gabriele Centi, Blanka Wichterlová, Alexis T. Bell, 2012-12-06 Atomically dispersed metal cations and small polyatomic cationic structures coordinated to the surface of porous matrices exhibit different properties from the same cationic species contained in a bulk oxide or supported on amorphous carriers This subject is treated to an extensive review showing how an understanding of it is essential to the development of a new generation of solid catalysts There are also exciting opportunities to shape the catalytic properties of the transition metal cations in microporous and mesoporous matrices The book covers both theoretical and experimental aspects including the distribution of framework Al atoms in Si rich zeolites distribution and siting of charge exchanged metal cations electronic adsorptive and catalytic properties of metal cations and correlation of metal cation structure and siting with catalytic activity

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