

Matching

Match each item with the correct statement below.

a. Boyle's law	d. Graham's law
b. Charles's law	e. Gay-Lussac's law
c. Dalton's law	f. ideal gas law (mark A & B)

- _____ 1) For a given mass of gas at constant temperature, the volume of the gas varies inversely with pressure.
- _____ 2) The volume of a fixed mass of gas is directly proportional to its Kelvin temperature, if the pressure is kept constant.
- _____ 3) The pressure of a gas is directly proportional to its Kelvin temperature if the volume is kept constant.
- _____ 4) $P \cdot V = n \cdot R \cdot T$
- _____ 5) At constant volume and temperature, the total pressure exerted by a mixture of gases is equal to the sum of the partial pressures of the component gases.
- _____ 6) The rate at which a gas will effuse is inversely proportional to the square root of the gas's molar mass.

Match each item with the correct statement below.

a. effusion	c. diffusion
b. compressibility	d. partial pressure

- _____ 7) a measure of how much the volume of matter decreases under pressure
- _____ 8) the pressure exerted by a gas in a mixture
- _____ 9) the escape of gas through a small hole in a container
- _____ 10) tendency of molecules to move to regions of lower concentration

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 11) Why is a gas easier to compress than a liquid or a solid?
 - a. Its volume increases more under pressure than an equal volume of liquid does.
 - b. Its volume increases more under pressure than an equal volume of solid does.
 - c. The space between gas particles is much less than the space between liquid or solid particles.
 - d. The volume of a gas's particles is small compared to the overall volume of the gas.
- _____ 12) Why does the pressure inside a container of gas increase if more gas is added to the container?
 - a. There is an increase in the number of collisions between particles and the walls of the container.
 - b. There is an increase in the temperature of the gas.
 - c. There is a decrease in the volume of the gas.
 - d. There is an increase in the force of the collisions between the particles and the walls of the container.
- _____ 13) How does the gas propellant move when an aerosol can is used?
 - a. from a region of high pressure to a region of lower pressure
 - b. from a region of high pressure to a region of equally high pressure
 - c. from a region of low pressure to a region of higher pressure

Chapter 14 Gases Study Guide Answer Key

Emilie Sanchez

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