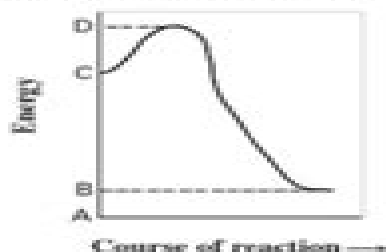


CHAPTER 17 REVIEW*Reaction Kinetics***SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. Refer to the energy diagram below to answer the following questions.



- d a. Which letter represents the energy of the activated complex?
- (a) A (c) C
(b) B (d) D
- c b. Which letter represents the energy of the reactants?
- (a) A (c) C
(b) B (d) D
- d c. Which of the following represents the quantity of activation energy for the forward reaction?
- (a) the amount of energy at C minus the amount of energy at B
(b) the amount of energy at D minus the amount of energy at A
(c) the amount of energy at D minus the amount of energy at B
(d) the amount of energy at D minus the amount of energy at C
- c d. Which of the following represents the quantity of activation energy for the reverse reaction?
- (a) the amount of energy at C minus the amount of energy at B
(b) the amount of energy at D minus the amount of energy at A
(c) the amount of energy at D minus the amount of energy at B
(d) the amount of energy at D minus the amount of energy at C
- b e. Which of the following represents the energy change for the forward reaction?
- (a) the amount of energy at C minus the amount of energy at B
(b) the amount of energy at B minus the amount of energy at C
(c) the amount of energy at D minus the amount of energy at B
(d) the amount of energy at B minus the amount of energy at A

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Harold Hart



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