

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence of a chemical reaction.

yes 1. A tomato smells rotten.

no 2. A drinking glass breaks into smaller pieces.

no 3. A piece of ice melts.

yes 4. Drain cleaner is mixed with water and the solution becomes warm.

yes 5. Candle wax burns.

yes/no 6. Molten candle wax solidifies.

yes 7. Green leaves turn yellow and red as the seasons change.

yes 8. Baking powder produces a gas that makes a cake rise.

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(g)	(g)
reactant	product	(g)	(g)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is also (9) reactant, oxygen is also (9) reactant, and water vapor is also (9) product. In a chemical equation for this reaction, also (12) arrow is used to separate hydrogen and oxygen from water vapor and energy. Also (13) plus sign is used to separate the symbols for hydrogen and oxygen. Also (14) (g) symbol is used to tell the state of hydrogen in the reaction, also (15) (g) symbol is used for the state of oxygen, and also (16) (g) symbol is used for the state of water vapor.

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Answers are included in the Appendix **Understanding Earth Student Study Guide** Peter L. Kresan, Reed Mencke, 2006-05-03 The guide helps students prepare for lectures and exams with a heavy emphasis on utilizing the book's Web resources **Study Guide and Full Solutions Manual** Susan McMurry, John McMurry, 2002-08 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all end of chapter problems in the textbook **Students Study Guide** Freddean, Pearson, 2003-07 **Chemistry Equations & Answers** Speedy Publishing, 2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to students even the best ones However a chemistry equations and answers study guide can help the student When used as a quick reference guide it can be used often to determine the formulas needed for various questions The astute student can cleverly devise ways to make the guide useful for test questions or other circumstances requiring one of the many chemistry equations

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