

# Chemistry Study Guide

# Periodic Table of the Elements

<small>1</small> <b>H</b> <small>1.008</small>																	<small>2</small> <b>He</b> <small>4.003</small>																															
<small>3</small> <b>Li</b> <small>6.941</small>	<small>4</small> <b>Be</b> <small>9.012</small>																	<small>5</small> <b>B</b> <small>10.81</small>	<small>6</small> <b>C</b> <small>12.01</small>	<small>7</small> <b>N</b> <small>14.01</small>	<small>8</small> <b>O</b> <small>15.99</small>	<small>9</small> <b>F</b> <small>18.99</small>	<small>10</small> <b>Ne</b> <small>20.18</small>																									
<small>11</small> <b>Na</b> <small>22.99</small>	<small>12</small> <b>Mg</b> <small>24.31</small>																	<small>13</small> <b>Al</b> <small>26.98</small>	<small>14</small> <b>Si</b> <small>28.09</small>	<small>15</small> <b>P</b> <small>30.97</small>	<small>16</small> <b>S</b> <small>32.06</small>	<small>17</small> <b>Cl</b> <small>35.45</small>	<small>18</small> <b>Ar</b> <small>39.95</small>																									
<small>19</small> <b>K</b> <small>39.09</small>	<small>20</small> <b>Ca</b> <small>40.08</small>	<small>21</small> <b>Sc</b> <small>44.96</small>	<small>22</small> <b>Ti</b> <small>47.88</small>	<small>23</small> <b>V</b> <small>50.94</small>	<small>24</small> <b>Cr</b> <small>51.99</small>	<small>25</small> <b>Mn</b> <small>54.94</small>	<small>26</small> <b>Fe</b> <small>55.85</small>	<small>27</small> <b>Co</b> <small>58.93</small>	<small>28</small> <b>Ni</b> <small>58.71</small>	<small>29</small> <b>Cu</b> <small>63.55</small>	<small>30</small> <b>Zn</b> <small>65.38</small>	<small>31</small> <b>Ga</b> <small>69.72</small>	<small>32</small> <b>Ge</b> <small>72.64</small>	<small>33</small> <b>As</b> <small>74.92</small>	<small>34</small> <b>Se</b> <small>78.96</small>	<small>35</small> <b>Br</b> <small>79.90</small>	<small>36</small> <b>Kr</b> <small>83.80</small>																															
<small>37</small> <b>Rb</b> <small>85.47</small>	<small>38</small> <b>Sr</b> <small>87.62</small>	<small>39</small> <b>Y</b> <small>88.91</small>	<small>40</small> <b>Zr</b> <small>91.22</small>	<small>41</small> <b>Nb</b> <small>92.91</small>	<small>42</small> <b>Mo</b> <small>95.94</small>	<small>43</small> <b>Tc</b> <small>98.00</small>	<small>44</small> <b>Ru</b> <small>101.07</small>	<small>45</small> <b>Rh</b> <small>102.91</small>	<small>46</small> <b>Pd</b> <small>106.42</small>	<small>47</small> <b>Ag</b> <small>107.87</small>	<small>48</small> <b>Cd</b> <small>112.41</small>	<small>49</small> <b>In</b> <small>114.82</small>	<small>50</small> <b>Sn</b> <small>118.71</small>	<small>51</small> <b>Sb</b> <small>121.76</small>	<small>52</small> <b>Te</b> <small>127.60</small>	<small>53</small> <b>I</b> <small>126.91</small>	<small>54</small> <b>Xe</b> <small>131.29</small>																															
<small>55</small> <b>Cs</b> <small>132.91</small>	<small>56</small> <b>Ba</b> <small>137.33</small>			<small>57</small> <b>Hf</b> <small>178.49</small>	<small>58</small> <b>Ta</b> <small>180.95</small>	<small>59</small> <b>W</b> <small>183.84</small>	<small>60</small> <b>Re</b> <small>186.21</small>	<small>61</small> <b>Os</b> <small>190.23</small>	<small>62</small> <b>Ir</b> <small>192.22</small>	<small>63</small> <b>Pt</b> <small>195.08</small>	<small>64</small> <b>Au</b> <small>196.97</small>	<small>65</small> <b>Hg</b> <small>200.59</small>	<small>66</small> <b>Tl</b> <small>204.38</small>	<small>67</small> <b>Pb</b> <small>207.2</small>	<small>68</small> <b>Bi</b> <small>208.98</small>	<small>69</small> <b>Po</b> <small>209</small>	<small>70</small> <b>At</b> <small>210</small>	<small>71</small> <b>Rn</b> <small>222</small>																														
<small>87</small> <b>Fr</b> <small>223</small>	<small>88</small> <b>Ra</b> <small>226</small>			<small>89</small> <b>Rf</b> <small>261</small>	<small>90</small> <b>Db</b> <small>262</small>	<small>91</small> <b>Sg</b> <small>266</small>	<small>92</small> <b>Bh</b> <small>264</small>	<small>93</small> <b>Hs</b> <small>277</small>	<small>94</small> <b>Mt</b> <small>268</small>	<small>95</small> <b>Ds</b> <small>271</small>	<small>96</small> <b>Rg</b> <small>272</small>	<small>97</small> <b>Cn</b> <small>285</small>	<small>98</small> <b>Uut</b> <small>289</small>	<small>99</small> <b>Fl</b> <small>288</small>	<small>100</small> <b>Uup</b> <small>294</small>	<small>101</small> <b>Lv</b> <small>293</small>	<small>102</small> <b>Uus</b> <small>294</small>	<small>103</small> <b>Uuo</b> <small>294</small>																														
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## Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

## Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

## Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons ( $e^-$ )
- **Inner part** → *nucleus* → houses the protons ( $p^+$ ) and neutrons ( $n$ )
- Outer & Inner part → together these are the **subatomic particles**

## Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	$p^+$	1	+1	Nucleus
Neutron	$n$	1	0	Nucleus
Electron	$e^-$	$\sim 0$	-1	Orbit/Shell

# Chemistry Study Guide For Oklahoma

**J Rink**



## **Chemistry Study Guide For Oklahoma:**

**Oklahoma Subject Area Tests Study Guide** Oklahoma Commission for Teacher Preparation, National Evaluation Systems, inc, 200? **OSAT Chemistry (104) Secrets Study Guide**, 2023 Study Guide for Oklahoma Subject Area Tests Oklahoma Commission for Teacher Preparation, 1990 Osat Chemistry (004) Secrets Study Guide: Ceoe Exam Review for the Certification Examinations for Oklahoma Educators / Oklahoma Subject Area Tests Ceoe Exam Secrets Test Prep, 2018-04-12 Includes Practice Test Questions OSAT Chemistry 004 Secrets helps you ace the Certification Examinations for Oklahoma Educators Oklahoma Subject Area Tests without weeks and months of endless studying Our comprehensive OSAT Chemistry 004 Secrets study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you've ever imagined OSAT Chemistry 004 Secrets includes The 5 Secret Keys to CEOE Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don't Procrastinate Test Yourself Introduction to the CEOE Series including CEOE Assessment Explanation Two Kinds of CEOE Assessments A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don't Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific CEOE exam and much more **Resources in Education**, 1984 **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office, 1979 *Atomic Energy Guidebook* Atomic Energy Guideletter, 1955 **Peterson's Guide to Graduate Programs in the Physical Sciences and Mathematics**, 1991 *Peterson's Annual Guides to Graduate Study*, 1983 **National Union Catalog**, 1973 Includes entries for maps and atlases **Resources in Education**, 1986 **Testing for Teacher Certification** Michael Chernoff, William P. Gorth, 1986 Teacher certification testing is the most widely discussed topic in education This book explores the history of the movement relevant legal and technical issues and program design and implementation Bibliographic Guide to Education, 1982 **The National Union Catalogs, 1963-**, 1964 *Bibliographic Guide to Government Publications* New York Public Library. Research Libraries, 1981 **The Science Teacher**, 1957 **Catalogue of High-school & College Textbooks, Including a Complete Index & Price List 1911** Ginn & co., publishers, Ginn and Company, 1911 *Peterson's Guide to Graduate Programs in Engineering and Applied Sciences*, 1991 **Peterson's Guide to Graduate Programs in the Biological and Agricultural Sciences**, 1990 **Peterson's Guide to Graduate Programs in the Biological Sciences 1997** Peterson's, 1997-01-05 Graduate students depend on this series and ask for it by name Why For over 30 years it's been the only one stop source that supplies all of their information needs The new

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## **Chemistry Study Guide For Oklahoma** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chemistry Study Guide For Oklahoma**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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