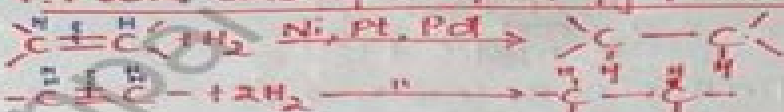


HYDROCARBON

MOP of Alkane

1. From unsaturated hydrocarbon



DM \rightarrow π bond ko H.H se tod do.

exothermic # ROR: Alkyne > Alkene

4 MCTS formed # ROR $\propto \frac{1}{s.H}$

Syn addition \propto stability

Addⁿ of conj. diene



RanNi \rightarrow alloy of Ni & Al in NaOH
Galvanom temp better than Ni (at 100°C)

If RanNi used \rightarrow Sabatier Senderason's

[Rh(PPh₃)₃Cl] used also

(Wilkinson's catalyst)

Alkene	Addition	Product
Cis	Syn	Erythro
Trans	Syn	Threo
Cis	Anti	Threo
Trans	Anti	Erythro

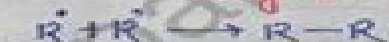
2. From alkyl halide



X halogen & lagao

ROR: R-Cl < R-Br < R-I ROR $\propto \frac{1}{BE}$

3. Wurtz reaction



produce higher alkane with even no. of 'C'

R-X \rightarrow coupling

(1,2)

R-X \rightarrow disproportionation

(3)

in a molecular wurtz faster than in a molecular but prod should be stable i.e. A.A nahi banana hai.

ROR $\propto \frac{1}{s.H}$ R-X > 2° R-X > 3° R-X

for same R: RF < RCl < RBr < RI

4. Decarboxylation reaction



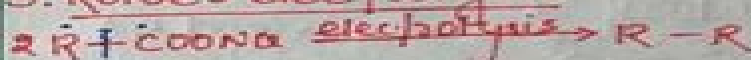
- COOH halogen & lagao



- COONa halogen & lagao

ROR \propto stability of intermediate (R⁻)

5. Kolbe's electrolysis



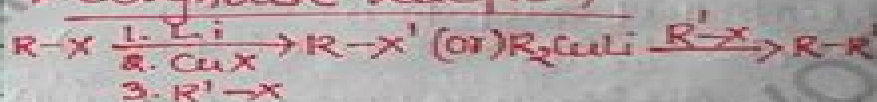
Cathode \rightarrow H₂ Anode \rightarrow Alkane & CO₂

Addition Reaction

6. Frankland reaction



7. Coreyhouse reaction



Attack of R⁻ \rightarrow RDS

ROR $\propto \frac{1}{s.H}$ of R'-X

If same R:

R-F < R-Cl < R-Br < R-I

BE \downarrow \rightarrow leaving tendency \uparrow

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Jill Meryl Levy



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