

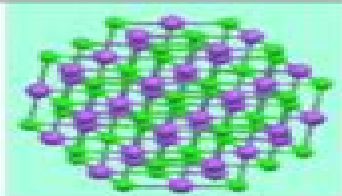
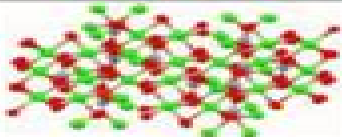

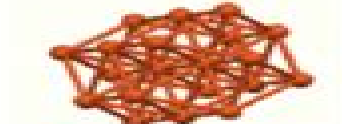

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## Unit 6 – Worksheet 2-key

### Why Structure is Important

In the table below, identify the type of solid (atomic, molecular, ionic) each substance forms. Then predict in which phase (solid, liquid or gas) each substance would exist on the Earth, on Mercury and on Pluto.

Substance	Crystal structure	Type	melting pt (°C) boiling pt (°C)	Phase at room temp on
NaCl		I	801 1413	Earth (25°C) <del>S</del> Mercury (450°C) <del>S</del> Pluto (-230°C) <del>S</del>
CaCO <sub>3</sub>		I	520 decomposes at 825	E <del>S</del> M <del>S</del> P <del>S</del>
S <sub>8</sub>		M	115 444	E <del>S</del> M G P <del>S</del>
Cu		A	1084 2567	E <del>S</del> M <del>S</del> P <del>S</del>
H <sub>2</sub> O <sub>2</sub>		M	-0.41 150	E L M G P <del>S</del>

# Chemistry Unit 6 Worksheet 2 Why Structure Is Important

**Robert C. Smoot, Smoot, Richard G.  
Smith, Jack Price**



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**Chemistry** James N. Spencer, George M. Bodner, Lyman H. Rickard, 2002-02-28 The second edition of Spencer's Chemistry Structure and Dynamics has been the most successful reform project published for the General Chemistry course The authors have revised the text by building on the recommendations of the ACS's Task Force on the General Chemistry Curriculum and suggestions from the adopters of the first edition This innovative text provides a fifteen chapter introduction to the fundamental concepts of Chemistry A collection of additional topics at the end of each chapter allow instructors to supplement and tailor their courses according to individual need Three major themes link the content of the book the process of science the relationship between molecular structure and physical chemical properties and the relationship between the microscopic and macroscopic levels

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