

Chemistry Study Guide

Remember that while this **CORRECTLY COMPLETED** study guide has a lot of valuable information that will be on the test, you are encouraged to also study notes in SJ, handouts, worksheets, and labs in order to best prepare for the test.

1. A characteristic of matter that can be seen or measured **WITHOUT CHEMICALLY CHANGING** that matter is a _____ property.
2. A characteristic of matter that can only be observed when one substance changes into or produces something new is a _____ property.
3. A change in which the identity of the substance does **NOT** change is known as a _____ change.
4. A change in which one type of matter transforms into another type or kind that may have very different properties from the original is known as a _____ change.
5. The state of matter in which the molecular configuration is the most spread out because the atoms have the most energy and are moving fast is the _____ state.
6. The state of matter that is created when heat (more energy) is added to a solid is the _____ state.
7. The state of matter with the least energy in which the atoms are tightly packed together and vibrate slowly in place is the _____ state.
8. A pure substance in its simplest form (a single atom) that cannot be broken down into any other substance by chemical or physical means is a(n) _____.
9. A pure substance made of two or more elements chemically combined in a set ratio is a _____.
10. When one or more of the same element combine and that substance can exist alone and keep its original properties is a _____.
11. In the chart below, put a check mark in the appropriate category for the symbol/formula listed. **Remember, all compounds are molecules, but not all molecules are compounds.**

Symbol/Formula	Atom (Element)	Molecule	Compound
NaCl			
C			
MgSO ₄			
O ₂			

Chem Study Guide

Matthew Bowling

Chem Study Guide:

Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples Introductory Chemistry Mike Lucarelli,David Alan Proctor,2002 Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series,Mark Kernion,Joseph A. Mascetta,2021-09-07 Barron's Science 360 Chemistry is your complete go to guide for everything chemistry This comprehensive guide is an essential resource for High school and college courses Homeschooling Virtual Learning Learning pods Inside you'll find Comprehensive Content Review Begin your study with the basic building block of chemistry and build as you go Topics include atomic structure chemical formulas electrochemistry the basics of organic chemistry and much more Effective Organization Topic organization and simple lesson formats break down the subject matter into manageable learning modules that help guide a successful study plan customized to your needs Clear Examples and Illustrations Easy to follow explanations hundreds of helpful illustrations and numerous step by step examples make this book ideal for self study and rapid learning Practice Exercises Each chapter ends with practice exercises designed to reinforce and extend key skills and concepts These checkup exercises along with the answers and solutions will help you assess your understanding and monitor your progress Access to Online Practice Take your learning online for 50 practice questions designed to test your knowledge with automated scoring to show you how far you have come **Chemistry Study Guide** CreateSpace Independent Publishing Platform,2014-09-09 Chemistry Study Guide for Students Who Struggle In Chemistry Inorganic

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