

# **Stoichiometry Practice Worksheet**

## **Balancing Equations and Simple Stoichiometry**

Balance the following equations:



Solve the following stoichiometry grams-grams problems:

6) Using the following equation:



How many grams of sodium sulfate will be formed if you start with 200 grams of sodium hydroxide and you have an excess of sulfuric acid?

7) Using the following equation:



How many grams of lithium nitrate will be needed to make 250 grams of lithium sulfate, assuming that you have an adequate amount of lead (IV) sulfate to do the reaction?

# Chemfiesta Stoichiometry Practice Answers

**Julie C. Gilbert**

## Chemfiesta Stoichiometry Practice Answers:

**Sassy Stoichiometry Problems** Julie C. Gilbert, 2021-03-14 Need more Stoichiometry practice? Stoichiometry has been striking fear into the hearts of chemistry students for ages. The best way to conquer something is to practice it! Inside you'll find Brief descriptions of each type of ideal stoichiometry and limiting reactant stoichiometry 4 ideal stoichiometry worksheets broken down by type with keys and explanations 4 ideal stoichiometry self quizzes with their answer keys 2 limiting reactant stoichiometry worksheets with keys and explanations 2 limiting reactant stoichiometry self quizzes with answer keys 2 mixed stoichiometry self tests with answer keys. This is a companion workbook for the 5 Steps to Surviving Chemistry book. However, you do not need to have read that book to find this workbook useful. [Chemistry Skill Builder](#) Steven D. Gammon, 1997 Chemistry SKILLBuilder gives students extra practice and feedback on three key topics: nomenclature, stoichiometry, and balancing equations. [Bite-Sized Chemistry Calculations](#) John Lambert, 2024-07-10 Bite-Sized Chemistry Calculations is a series of books on chemistry calculations aimed at helping students overcome the challenges associated with tackling the various types of calculations encountered in different aspects of chemistry, focusing on a few topics at a time to facilitate comprehension. Written by an experienced chemistry educator, each book in the series has been tailored to fully meet the needs of students at all levels, especially those taking college level general chemistry courses as well as those following various O level curricula worldwide. This part of the series explores the different types of problems and calculations encountered in mass, the mole, and stoichiometry, including the determination of formulae of ionic compounds, relative formula masses, mass and percent compositions of compounds, all aspects of mole calculations, empirical and molecular formulae calculations based on chemical equations, limiting reagents, gas stoichiometry, and percent yield. The series is packed with many salient features that are meant to facilitate both teaching and learning. Some of these include helpful explanations, many examples, alternative ways to solve problems, plenty of practice questions, complete answers, and appendices. With this book, you will be well prepared for your exams and boost your performance. CONTENTS: 1 Writing the Formulae of Ionic Compounds, 2 Formula Masses, 3 Mass and Percent Compositions, 4 The Mole and Mass, 5 The Mole and Number of Particles, 6 The Mole and Concentration, 7 The Mole and Molar Volume, 8 Empirical and Molecular Formulae, 9 Chemical Equations, 10 Calculations Based on Chemical Equations, 11 Limiting Reagent, 12 Gas Volume, Volume Stoichiometry, 13 Percent Yield, Answers to Practice Problems, Appendices.

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