

# CHEMICAL REACTIONS

## webquest

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**CHEMICAL REACTIONS** Name: \_\_\_\_\_

Subplan about chemical reactions by visiting the link below:  
[http://www.ck12.org/chemistry/chemical-reactions/](#)

1. What happens to matter in a chemical reaction?
2. Write the equation for a chemical reaction? Explain.
3. What happens to the reactants and products in a reaction?
4. Give the examples of the reaction that occurring at a school in following in the diagram.
5. What do you notice about the number of hydrogen atoms and oxygen atoms in each side of the equation?
6. Complete the chart below with the four signs of chemical reactions.

TYPE OF EVIDENCE	REACTIONS

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**CHEMICAL REACTIONS** Name: \_\_\_\_\_

Subplan about the equations of chemical reactions by visiting the link below:  
[http://www.ck12.org/chemistry/chemical-reactions/](#)

1. What does the law of conservation of mass tell us?
2. Chemical reactions are all about \_\_\_\_\_ atoms.
3. In the chemical equation (2H<sub>2</sub> + O<sub>2</sub> → 2H<sub>2</sub>O), how many oxygen atoms are there? How do you get to this number?
4. In the chemical equation (2H<sub>2</sub> + O<sub>2</sub> → 2H<sub>2</sub>O), how many oxygen atoms are there? How do you get to this number?
5. Complete the chart with the number of oxygen atoms in each chemical equation. Please the table and try to understand on your own. Then, use the table to make your answers.

Chemical Equation	Reactants	Products	Oxygen Atoms	Hydrogen Atoms
2H <sub>2</sub> + O <sub>2</sub> → 2H <sub>2</sub> O				

Complete with balancing chemical equations in part 2 of the video series by visiting below:  
[http://www.ck12.org/chemistry/chemical-reactions/](#)

6. Remember, when balancing the equation, change the number \_\_\_\_\_ of compounds.
7. Please the table and try to balance the equations below. Use the charts to help you!

Chemical Equation	Reactants	Products	Oxygen Atoms	Hydrogen Atoms
2H <sub>2</sub> + O <sub>2</sub> → 2H <sub>2</sub> O				

Chemical Equation	Reactants	Products	Oxygen Atoms	Hydrogen Atoms
2H <sub>2</sub> + O <sub>2</sub> → 2H <sub>2</sub> O				

Try to balance the equation below:  
2H<sub>2</sub> + O<sub>2</sub> → 2H<sub>2</sub>O



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# Chemical Reaction Webquest

**Rachael Morlock**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Conference proceedings. New perspectives in science education Pixel, 2014 *Chemical Reactions* Kristi Lew, 2009 Explores how chemical reactions happen from how your body breaks down the food you eat to how forensic experts use them to solve crimes **British Education Index**, 1999 **The Chemical News and Journal of Industrial Science**, 1928

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leading researchers from eight countries across four continents this book is an important compendium of syntheses and insights for veteran researchers teachers and curriculum designers eager to advance the SSI agenda **State Education Journal Index** ,2006 Konkrete Fachdidaktik Chemie Katrin Sommer,Judith Wambach-Laicher,Peter Pfeifer,2025-01-01 Das Standardwerk Konkrete Fachdidaktik Chemie liegt in einer grundlegend neu bearbeiteten Fassung vor in der es den Anforderungen eines erfolgreichen kompetenzorientierten Chemieunterrichts gerecht wird Zum ersten Mal erarbeiten renommierte Autorinnen und Autoren der ersten und zweiten Phase der Lehrerbildung gemeinsam ein chemiedidaktisches Lehrwerk Damit liegt ein universelles Lehrbuch f r alle Phasen der Lehrerbildung vor das den spezifischen Bed rfnissen und Sichtweisen des Chemieunterrichts Rechnung tr gt F r Studium Referendariat und Schulpraxis bietet es einen Blick auf den gesamten Chemieunterricht ein reiches Anregungspotential f r didaktisch fundierte methodische Ma nahmen und Entscheidungen Argumente und Anregungen dass guter lernwirksamer Chemieunterricht in seinen Grundstrukturen planbar ist zugleich aber offen bleibt f r einen sensiblen und engagierten Umgang der Lehrenden mit den Sch lerinnen und Sch lern in fruchtbaren Situationen des Unterrichtsgeschehens Das Lehrwerk richtet sich an alle Lehramtsstudierende Referendare und Lehrkr fte mit Chemie als Unterrichtsfach in allen Schulformen der Sekundarstufe I und II *The Basics of Chemical Reactions* Krista West,2013-12-15 Explores the simplicity of basic chemical reactions and then builds to the more complex giving readers a history of the years and the minds that contributed to the research that led to chemistry as we know it today **Secondary Teachers Guide to Free Curriculum Materials** ,2010 **Chemical Reactions** Denise Walker,2007 This title introduces the reader to the huge variety of chemical reactions that shape our world Find out all about explosions learn about how to start reactions and understand how chemical equations work Chemical Reactions Philip Wolny,2011-01-15 Learn about chemical reactions what they are the people responsible for helping us understand them and how they affect us in the world today **Chemical Reactions: It Matters** Rachael Morlock,2019-12-15 What do a rusting bike a growling stomach and a crackling campfire have in common They are all the result of chemical reactions Chemicals are constantly reacting in our bodies and all around us Chemical reactions can be natural like respiration and photosynthesis or human made like the reactions in fertilizer and fireworks This book simplifies chemical reactions with clear explanations and explosive examples Primary sources show how scientists and engineers have studied and harnessed the power of chemical reactions Their bold achievements will inspire readers to look for the extraordinary ways chemical reactions shape our world and future *Chemical Reactions* Jacqueline Barber,1998 An ordinary sandwich bag becomes a safe laboratory as students mix chemicals that bubble change color and produce gas heat and odor Students then experiment to determine what causes the heat in this chemical reaction Fizz, Gurgle, Pop! Chemical Reactions in the Lab Daniel R. Faust,2020-07-15 Whether they are testing a new drug creating new materials or checking the safety of our food and water all scientists follow the scientific method One of the major steps in the scientific method is testing your hypothesis That s when scientists hit the lab

This volume stresses the importance of safety in the laboratory whether it is a school science lab or the most high tech research facility. It also examines the proper rules and procedures everyone must follow when working with chemicals in the lab. Fact boxes and full color photographs support the age appropriate text. **Chemical Reactions** Jenny Karpelenia, 2006

Learn about chemical reactions including what causes them and the elements and compounds that react with one another to achieve stability within the reaction. **Super Cool Chemical Reaction Activities with Max Axiom** Agnieszka Biskup, 2015-04-09

From bouncing eggs to monster toothpaste to colour changing liquids Super Scientist Max Axiom makes hands on science experiments super cool. From the secret files of Axiom Laboratories comes a collection of kid friendly chemical reaction experiments and activities. With clear step by step photo illustrated instructions these projects will not only teach core curriculum science concepts but also amaze aspiring young scientists. **Chemical Reactions** Louise Spilsbury, Richard Spilsbury, 2007

Readers will learn what chemical reactions are how they work what changes happen during reactions and how we can stop reactions. *Encyclopedia of Chemical Reactions* C. A. Jacobson, 1946

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