

molecular equation:



complete ionic equation:



net ionic equation:



spectator ions: Cl^- , Na^+

general reaction type: Double replacement / displacement

VI. Heating Cu with Atmospheric O₂

appearance of the unreacted Cu mesh:

Shiny copper

appearance of the Cu mesh after heating and cooling:

dull dark brown



general reaction type: Combination

Which species are oxidized? Cu

Which species are reduced? O

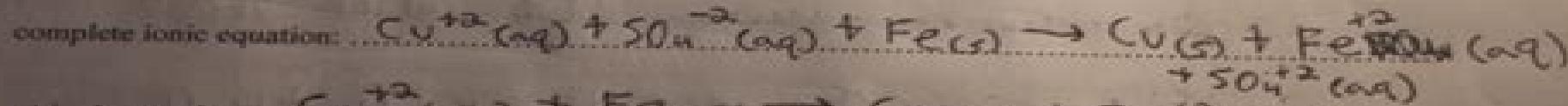
VII. Reacting CuSO₄ Solution with Steel Wool (Fe)

appearance of the steel wool (Fe) before reaction: silver solid

appearance of the CuSO₄ solution before reaction: light blue

appearance of the steel wool (Fe) after the reaction has occurred: dark brown solid

appearance of the CuSO₄ solution after the reaction has occurred: light blue liquid



spectator ions: SO_4^{-2} (aq) general reaction type: Single Displacement

Which species are oxidized? Fe

Which species are reduced? Cu⁺²

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Ralph Thomas Myers

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