

Balancing and Reactions Test Unit 2 SCH3U**Name:**

 K T C A **Assessment of Learning (Product)**
15 20 5 4

Multiple Choice: Circle the most correct answer. (10K)

1. Propane gas, C_3H_8 , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for H_2O is...

- A) 1 B) 2 C) 3 D) 4 E) 5 f) 6

2. State the type of reaction: $2 Na + 2 H_2O \rightarrow 2 NaOH + H_2$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

3. Which of the following types of chemical reactions produces water and an ionic compound?

- A) single displacement B) neutralization C) decomposition D) synthesis

4. State the type of reaction: $Ca(NO_3)_2 + 2 HCl \rightarrow CaCl_2 + 2 HNO_3$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

5. If chlorine gas is produced by single displacement, the other halogen in the reaction must be

- A) bromine B) iodine C) astatine D) fluorine

6. In what kind of reaction do two or more substances combine to form a new compound?

- A) decomposition reaction C) double-displacement reaction
B) single displacement D) synthesis reaction

7. A group whose atoms (M) undergo the reaction:



- A) group 1 B) group 2 C) group 13 D) group 14 E) group 15

8. When heated, a metal carbonate decomposes into a metal oxide and

- A) carbon. B) carbon dioxide C) oxygen. D) hydrogen.

9. Oxides of active metals, such as CaO , react with water to produce

- A) metal carbonates. B) metal hydrides. C) acids. D) metal hydroxides.

10. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals. C) Two reactants produce two products.
B) They involve a single product. D) Any metal replaces any other metal.

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this book was to create a textbook for students a textbook that presents the material in a way that they learn to solve all the questions along with the strategy to approach the problems In this book we mixed all our teaching experience of 15 years along with theoretical and experimental knowledge to generate a hand book for all students to reason their way to a solution rather than memorize a multitude of facts hoping they don t run out of memory This book covers mainly 6 units with 59

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Table of Contents Chemical Reactions Test A Answers

1. Understanding the eBook Chemical Reactions Test A Answers
 - The Rise of Digital Reading Chemical Reactions Test A Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Reactions Test A Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Reactions Test A Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Reactions Test A Answers
 - Personalized Recommendations
 - Chemical Reactions Test A Answers User Reviews and Ratings
 - Chemical Reactions Test A Answers and Bestseller Lists
5. Accessing Chemical Reactions Test A Answers Free and Paid eBooks
 - Chemical Reactions Test A Answers Public Domain eBooks
 - Chemical Reactions Test A Answers eBook Subscription Services
 - Chemical Reactions Test A Answers Budget-Friendly Options

6. Navigating Chemical Reactions Test A Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Reactions Test A Answers Compatibility with Devices
 - Chemical Reactions Test A Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Reactions Test A Answers
 - Highlighting and Note-Taking Chemical Reactions Test A Answers
 - Interactive Elements Chemical Reactions Test A Answers
8. Staying Engaged with Chemical Reactions Test A Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Reactions Test A Answers
9. Balancing eBooks and Physical Books Chemical Reactions Test A Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Reactions Test A Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Reactions Test A Answers
 - Setting Reading Goals Chemical Reactions Test A Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Reactions Test A Answers
 - Fact-Checking eBook Content of Chemical Reactions Test A Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

-
- Interactive and Gamified eBooks

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