

# Chemistry Review Packet

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## A. Chemistry of the following reactions:



Which 0.5 moles of hydrogen are released with 0.500 moles of oxygen. How many grams of water are released?

- A. 10.00 g water
- B. 20.00 g water
- C. 30.00 g water
- D. 40.00 g water
- E. 50.00 g water

12. Which of the following is a 1.0 M aqueous solution? Use the following pKa's:

- A.  $\text{H}_2\text{SO}_4$
- B.  $\text{H}_2\text{SO}_3$
- C.  $\text{H}_2\text{SO}_3$
- D.  $\text{H}_2\text{SO}_4$
- E.  $\text{H}_2\text{SO}_3$

13. Which of the following substances would be broken down into simpler substances by physical or chemical means?

- A. Water
- B. Baking soda
- C. Bleach
- D. Peroxide
- E. Water

14. What is the approximate percent mass position by mass of hydrogen in the compound ethyl,  $\text{C}_2\text{H}_5\text{OH}$  (propanol) weight, in percent?  $(\text{C} = 12.00\text{ g/mol} \times 2.00\text{ g/mol}) / (\text{H} = 1.00\text{ g/mol} \times 6.00\text{ g/mol})$

- A. 10.00%
- B. 10.00%
- C. 20.00%
- D. 24.00%
- E. 25.00%

15. The empirical formula of a certain hydrocarbon is  $\text{C}_2\text{H}_6$ , and it has a molecular weight equal to 60.00. Based on this information, the molecular formula of this hydrocarbon is:

- A.  $\text{C}_2\text{H}_6$
- B.  $\text{C}_2\text{H}_{12}$
- C.  $\text{C}_2\text{H}_6$
- D.  $\text{C}_2\text{H}_6$
- E.  $\text{C}_2\text{H}_6$

16. Which of the following aqueous solutions is 1.0 Molar? Use the following percent solutions: percent?

- A. 10.00%  $\text{CH}_3\text{CO}_2\text{H}$
- B. 10.00%  $\text{H}_2\text{SO}_4$
- C. 10.00%  $\text{H}_2\text{SO}_3$
- D. 10.00%  $\text{H}_2\text{SO}_4$
- E. 10.00%  $\text{H}_2\text{SO}_3$

## B. Chemistry of the reactions:



What is this reaction best classified?

- A. Precipitation
- B. Redox reaction
- C. Single replacement
- D. Combination
- E. Hydratation

17. The following reaction tells you which order of reactivity and of reagents:



Which of the following reagents would you choose to react with iron,  $\text{Fe}$ , to form  $\text{FeCl}_3$  (iron(III) chloride)?

- A. 1.00 M  $\text{H}_2\text{O}$
- B. 1.00 M  $\text{H}_2\text{O}$
- C. 1.00 M  $\text{H}_2\text{O}$
- D. 1.00 M  $\text{H}_2\text{O}$
- E. 1.00 M  $\text{H}_2\text{O}$

# Chemistry 1 Review Packet

**Ahmed H Zewail**

## Chemistry 1 Review Packet:

Femtochemistry Ahmed H. Zewail, 1994 These two volumes on Femtochemistry present a timely contribution to a field central to the understanding of the dynamics of the chemical bond. This century has witnessed great strides in time and space resolutions down to the atomic scale providing chemists, biologists, and physicists with unprecedented opportunities for seeing microscopic structures and dynamics. Femtochemistry is concerned with the time resolution of the most elementary motions of atoms during chemical change bond breaking and bond making on the femtosecond 10-15 second time scale. This atomic scale of time resolution has now reached the ultimate for the chemical bond and as Lord George Porter puts it chemists are near the end of the race against time. These two volumes cover the general concepts, techniques, and applications of femtochemistry. Professor Ahmed Zewail who has made the pioneering contributions in this field has from over 250 publications selected the articles for this anthology. These volumes begin with a commentary and a historical chronology of the milestones. He then presents a broad perspective of the current state of knowledge in femtochemistry by researchers around the world and discusses possible new directions. In the words of a colleague it is a must on the reading list for all of my students all readers will find this to be an informative and valuable overview. The introductory articles in Volume I provide reviews for both the non-experts as well as for experts in the field. This is followed by papers on the basic concepts. For applications elementary reactions are studied first and then complex reactions. Volume I is complete with studies of solvation dynamics, non-reactive systems, ultrafast electron diffraction, and the control of chemical reactions. Volume II continues with reaction rates, the concept of elementary intramolecular vibrational energy redistribution (IVR) and the phenomena of rotational coherence which has become a powerful tool for the determination of molecular structure via time resolution. The second volume ends with an extensive list of references according to topics based on work by Professor Zewail and his group at Caltech. These collected works by Professor Zewail will certainly be indispensable to both experts and beginners in the field. The author is known for his clarity and for his creative and systematic contributions. These volumes will be of interest and should prove useful to chemists, biologists, and physicists. As noted by Professor J. Manz Berlin and Professor A. W. Castleman Jr. Penn State femtochemistry is yielding exciting new discoveries from analysis to control of chemical reactions with applications in many domains of chemistry and related fields e.g. physical organic and inorganic chemistry, surface science, molecular biology, etc.

*Femtochemistry: Ultrafast Dynamics Of The Chemical Bond (In 2 Volumes)* -

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atomic scale of time resolution has now reached the ultimate for the chemical bond and as Lord George Porter puts it chemists are near the end of the race against time These two volumes cover the general concepts techniques and applications of femtochemistry Professor Ahmed Zewail who has made the pioneering contributions in this field has from over 250 publications selected the articles for this anthology These volumes begin with a commentary and a historical chronology of the milestones He then presents a broad perspective of the current state of knowledge in femtochemistry by researchers around the world and discusses possible new directions In the words of a colleague it is a must on the reading list for all of my students all readers will find this to be an informative and valuable overview The introductory articles in Volume I provide reviews for both the non experts as well as for experts in the field This is followed by papers on the basic concepts For applications elementary reactions are studied first and then complex reactions Volume I is complete with studies of solvation dynamics non reactive systems ultrafast electron diffraction and the control of chemical reactions Volume II continues with reaction rates the concept of elementary intramolecular vibrational energy redistribution IVR and the phenomena of rotational coherence which has become a powerful tool for the determination of molecular structure via time resolution The second volume ends with an extensive list of references according to topics based on work by Professor Zewail and his group at Caltech These collected works by Professor Zewail will certainly be indispensable to both experts and beginners in the field The author is known for his clarity and for his creative and systematic contributions These volumes will be of interest and should prove useful to chemists biologists and physicists As noted by Professor J Manz Berlin and Professor A W Castleman Jr Penn State femtochemistry is yielding exciting new discoveries from analysis to control of chemical reactions with applications in many domains of chemistry and related fields e g physical organic and inorganic chemistry surface science molecular biology etc

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Thomas Weinacht,Brett J. Pearson,2018-12-21 This concise and carefully developed text offers a reader friendly guide to the basics of time resolved spectroscopy with an emphasis on experimental implementation The authors carefully explain and relate for the reader how measurements are connected to the core physical principles They use the time dependent wave packet as a building block for understanding quantum dynamics progressively advancing to more complex topics The topics are discussed in paired sections one discussing the theory and the next presenting the related experimental methods A wide range of readers including students and newcomers to the field will gain a clear and practical understanding of how to measure aspects of molecular dynamics such as wave packet motion intramolecular vibrational relaxation and electron electron coupling and how to describe such measurements mathematically

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Gardener's Monthly and Horticultural Advertiser ,1879      **Health** Frances Sienkiewicz Sizer,2000 This student textbook for grades 9 12 covers emotional and physical health topics along with drug abuse information human reproduction general fitness and nutrition and the environment and health impact each other

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