

## Study Guide: Honors Chemistry Midterm 2025

Answer the following questions in separate sheets of paper. For answers requiring explanation, explain! For calculations, show all work! This study guide is graded!

### \*\*\*Dates of the midterm\*\*\*

A Days: Wednesday, January 8  
B Days: Friday, January 10

### \*\*\*Dates of midterm corrections\*\*\*

A Days: Monday, January 13  
B Days: Tuesday, January 14

### Physical and Chemical Properties

- 1) All matter has what two characteristics (that distinguish it from energy)?  
Have mass and takes up space (volume).
- 2) Intensive physical property versus extensive physical property
  - a. What is the difference between the two?  
Intensive properties do not depend on the amount of substance present and extensive properties depend on the quantity or the amount of substance.
  - b. Give two examples of each.  
Density and boiling point for Intensive.  
Mass and volume for Extensive.
  - c. Explain why only intensive physical properties can be used to identify an unknown object.  
Intensive properties are unique to a material and remain constant regardless of how much of the material is present. In contrast, extensive properties like mass and volume depend on the sample size and cannot uniquely identify a material.
- 3) What is the difference between a physical change and a chemical change?  
A physical change affects the substance's appearance, state, or properties, but its chemical composition remains unchanged and chemical change involves a reaction that alters the substance's chemical composition, resulting in the formation of new substances with different properties.
- 4) Label each of these as Physical (P) or Chemical (C) changes:
  - A. Food going bad (C)
  - B. Gold melting (P)
  - C. Salt dissolving (P)
  - D. Wood burning (C)
  - E. Paper tearing (P)
  - F. Water vaporizing (P)
- 5) Density
  - a. What is density?  
Density is a physical that measures how much mass is contained in each volume.

# Chemistry Honors Ceoce Study Guide Answers

**Steven S. Zumdahl,Donald J. DeCoste**

## **Chemistry Honors Ceoce Study Guide Answers:**

*Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry* Denise Guinn,Rebecca Brewer,Rachel C. Lum,2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text *Study Guide for Chemical Principles [by] Steven S. Zumdahl* Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples *Student Study Guide for Chemistry* John E. McMurry,Robert C. Fay,Jill Kirsten Robinson,Julie Klare,Donnajean Fredeen,James Zubricky,2015-03-09 The Study Guide includes learning goals an overview a review section with worked examples and self tests with answers

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