

General Chemistry II Study Guide

Questions	Answers
At what speed do molecules move?	Molecules move at a variety of speeds depending on the amount of energy in them.
What could cause a molecule to escape the IMF in which it is held?	A large amount of energy would cause a molecule to escape the liquid that it is in.
What is vapor pressure?	The pressure (concentration) of gas molecules above the liquid.
Which molecules will have vapor pressure?	All molecules have some sort of vapor pressure
What is the relationship between IMFs and Pvap?	As the strength of the IMF increases the Pvap decreases.
What is the relationship between temperature and Pvap?	As temperature increases, Pvap also increases.
What is the definition of boiling?	When thermal energy is greater than the IMF
What is the real definition of boiling point?	When enough thermal energy has been added to give all the molecules the ability to escape.
What is the external pressure equal to?	1 atm
What does atm stand for?	atmosphere
If there are two jars, one with hydrogen bonding and one with dipole-dipole, which would you expect to have a higher vapor pressure?	Dipole-dipole
When a substance goes from a solid to	Sublimation

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Mary H. Scanlan

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1. Understanding the eBook Chemistry Ii Study Guide
 - The Rise of Digital Reading Chemistry Ii Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Ii Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Ii Study Guide
 - User-Friendly Interface
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 - Personalized Recommendations

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- Chemistry Ii Study Guide Enhanced eBook Features

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- Highlighting and Note-Taking Chemistry Ii Study Guide
- Interactive Elements Chemistry Ii Study Guide

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- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry Ii Study Guide

9. Balancing eBooks and Physical Books Chemistry Ii Study Guide

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Chemistry Ii Study Guide

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Chemistry Ii Study Guide

- Setting Reading Goals Chemistry Ii Study Guide
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry Ii Study Guide

- Fact-Checking eBook Content of Chemistry Ii Study Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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