



## Introduction to working as a Chemist.

Use a highlighter as you work through this booklet. ESL students, make vocab lists of the terminology that is defined or described here.

## Assessment Statements

- Outline the Scientific Method**
- Identify independent, dependent & controlled variables.**
- Demonstrate correct and safe use of laboratory equipment.**
- Distinguish between quantitative and qualitative data**
  - > Qualitative is descriptive, not using numbers, e.g. properties, colour, gas production, sound, odor.
  - > Quantitative is mathematical, including
    - Raw data (as recorded)
    - Processed (after calculations)
- Distinguish between accuracy and precision**
  - > Accuracy describes the correctness of a set of measurements
  - > Precision describes the repeatability of a set of measurements
- State SI units of measurement for**
  - > Mass: grams (g)
  - > Length: Metres (m), millimeters (mm)
  - > Concentration: molarity (M)
  - > Temperature: Degrees Celsius (°C)
  - > Volume: Litres (L) or millilitres (mL)
  - > Density: grams per cubic centimeter (g/cm<sup>3</sup>)
- Conductivity: Siemens (S)**
- Acidity or alkalinity (pH)**
- Energy: Joules (J or kJ)**
  - Determine the uncertainty of digital and analog measuring tools
  - > Usually  $\pm 1$  of the smallest division
- Calculate the % error of a given or recorded set of values**
  - >  $\frac{(\text{Measured} - \text{True})}{\text{True}} \times 100$
- State values to appropriate numbers of significant figures**
- Calculate values to appropriate numbers of significant figures**

## Related Concepts:

- 01.1 Element symbols** (Quia quiz: <http://www.quia.com/quiz/3246640.html>)
- 01.2 Accuracy, Precision, Error, Uncertainty** (Quia quiz: <http://www.quia.com/quiz/3507626.html>)
- 01.3 SigDigs and Notation** (Quia quiz: <http://www.quia.com/quiz/3504378.html>)

## Command Terms: Be sure you understand what is being asked of you

Analyse Break down to identify key parts. Interpret information to reach conclusions.

<b>Apply</b>	Use knowledge & understanding in response to a real situation or given circumstances.
<b>Calculate</b>	Obtain a numerical answer, showing relevant stages in the working.
<b>Define</b>	Give the precise meaning of a word, phrase, concept or physical quantity.
<b>Describe</b>	Give a detailed account or picture of an event, pattern, process or pattern.
<b>Design</b>	Produce a plan, simulation or model.
<b>Determine</b>	Obtain the only possible answer.
<b>Discuss</b>	Offer a considered and balanced review that includes a range of arguments, factors or hypotheses. Opinions or conclusions should be presented clearly and supported by appropriate evidence.
<b>Distinguish</b>	Make clear the differences between two or more concepts or items.
<b>Evaluate</b>	Assess the implications and limitations; make judgments about the ideas, works, solutions or ideas in relation to selected criteria.
<b>Explain</b>	Give a detailed account including reasons or causes.
<b>Identify</b>	Provide an answer from a number of possibilities. Recognise and state briefly a distinguishing fact or feature.
<b>Label</b>	Add title, labels or brief explanation(s) to a diagram or graph.
<b>List</b>	Give a sequence of answers with no explanation.
<b>Outline</b>	Give a brief account.
<b>Recall</b>	Remember or recognize from prior learning experiences.
<b>State</b>	Give a specific name, value or other brief answer without explanation or calculation.
<b>Suggest</b>	Propose a solution, hypothesis or other possible answer.

# Chemistry Lab Manual Solution Ubc

**Judd E. Hollander**

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