

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about *physical properties* and *chemical properties* of matter.

Use each of the terms below just once to complete the passage.

chemical
density

mass
properties

physical
substance

Matter is anything with (1) _____ and volume. A (2) _____ is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) _____ that can be observed. Substances have both physical and chemical properties. (4) _____ properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) _____ are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) _____ properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

_____ 7. Chemical formula H_2O
_____ 8. Forms green carbonate when exposed to moist air
_____ 9. Remains unchanged when in the presence of nitrogen
_____ 10. Colorless
_____ 11. Solid at normal temperatures and pressures
_____ 12. Ability to combine with another substance
_____ 13. Melting point
_____ 14. Liquid at normal temperatures and pressures
_____ 15. Boiling point is 100°C
_____ 16. Conducts electricity
_____ 17. Density is $\frac{1\text{g}}{\text{cm}^3}$

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