

**Chapter 2**  
**Chemistry of Life Study Guide**

1. Chemical reactions that absorb more energy than they release are called \_\_\_\_\_.
2. The first carbon compounds that scientists studied are called \_\_\_\_\_ compounds because they came from living organisms. (This word means it contains "carbon")
3. DNA and RNA are two types of \_\_\_\_\_. (type of macromolecule)
4. What is the process that changes one set of chemicals into another set of chemicals?
5. When salt is dissolved in water, water is the \_\_\_\_\_.
6. Enzymes affect chemical reactions in living organisms by \_\_\_\_\_ the activation energy of a reaction.
7. What atoms make up a molecule of water? How many hydrogen and oxygen?
8. The three particles that make up an atom are called protons, \_\_\_\_\_, and \_\_\_\_\_.
9. What type of ion forms when an atom loses electrons? (positive/negative)
10. What are the four main groups of carbon-based molecules (macromolecules)?
11. The activation energy needed for a chemical reaction is decreased by a \_\_\_\_\_. (another name for an enzyme)
12. \_\_\_\_\_ is a term that describes a substance formed by the combination of two or more elements in a definite proportion.
13. The most abundant compound in most living things is \_\_\_\_\_.
14. What category of carbon-based molecules (macromolecules) includes sugars and starches?
15. Fats, oils, and cholesterol are all types of \_\_\_\_\_.
16. Identify the reactants in the following chemical equation:



17. Chemical reactions change substances into different substances by \_\_\_\_\_ and chemical bonds.
18. A very strong base might have a pH of \_\_\_\_\_.

# Chemistry Of Life Study Guide With Answers

**Martha R. Taylor**



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