

# Kingsway Regional Middle School Chemical Reactions Science Study Guide

## Vocab

**Chemical Reaction:** The process by which one or more substances change to produce one or more different substances.

**Reactant:** A substance that takes part in and undergoes a change during a chemical reaction.



**Product:** A substance that is produced during a chemical reaction.



**Coefficient:** A number placed in front of a symbol or formula in a chemical equation. It shows how many of that substance are in the chemical reaction, acting like a multiplier to balance the equation.



**Subscript:** A number written below and to the right of a chemical symbol in a chemical formula. It shows how many atoms are in that molecule.



**Law of Conservation of Matter/Mass:** The Law of Conservation of Matter states that matter cannot be created or destroyed, but simply rearranged.

**Chemical Equation:** Chemical formulas and symbols written to represent a reaction.

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**Steven S. Zumdahl, Paul B. Kelter**



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