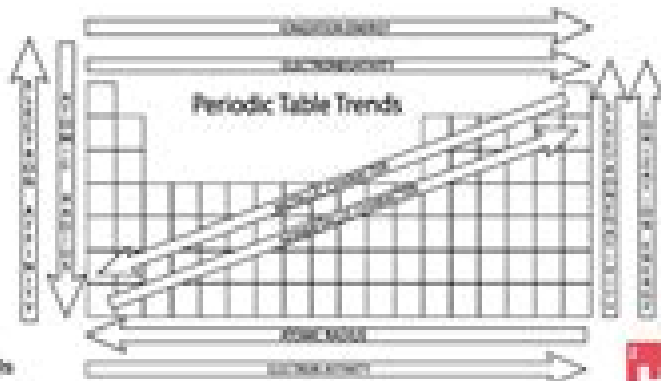


PERIODIC TABLE OF THE ELEMENTS

Atomic Number → **1** ← Symbol
Name → **Hydrogen** ← Atomic Weight

-  Solid
 Liquid
 Gas
- *At room temperature

[illegible]

Electron Configuration Effects

The diagram illustrates the distribution of atomic orbitals across the periodic table. The s-block (purple) includes groups 1 and 2. The p-block (orange) includes groups 13 through 18. The d-block (green) includes groups 3 through 10. The f-block (dark green) includes the lanthanide and actinide series. The orbitals are labeled as follows:

- s-block:** s
- p-block:** p
- d-block:** d
- f-block:** f

The lanthanide series (La to Lu) and actinide series (Ac to Lr) are shown below the main table, with their respective orbital configurations.

Chemistry Guide

Daniel Slosberg



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