

Chemistry Practice Test: 20 Ready-States (20, 21, 22, 23) & Free-Response Items

20. A student places a sample of copper oxide and a sample of aluminum oxide in separate crucibles. After heating, the student measures the mass of each crucible.

a. If the student's data are as follows, calculate the percent aluminum oxide in the sample.

Sample	Initial Mass	Final Mass
Copper Oxide	1.00 g	0.85 g
Aluminum Oxide	1.00 g	0.95 g

b. Calculate the percent error in the student's data.

c. Calculate the percent error in the student's data.

Answers

21. A student weighs four pieces of aluminum foil and finds the following data:
Aluminum foil mass (g) 1.00 1.00 1.00 1.00
Aluminum foil area (cm²) 1.00 1.00 1.00 1.00

a. If the student's data are as follows, calculate the percent error in the student's data.

$$\begin{aligned} \text{Aluminum foil mass (g)} &= \frac{1.00}{4} \text{ g} \\ \text{Aluminum foil area (cm}^2\text{)} &= \frac{1.00}{4} \text{ cm}^2 \\ &= 0.25 \text{ cm}^2 \end{aligned}$$

b. If the sample is placed in a graduated cylinder, the volume using 3 significant figures is

$$12.5 \text{ mL}$$

c. If the sample is placed in a graduated cylinder, the volume using 3 significant figures is

a. 12.5 mL

b. 12.5 mL

$$\begin{aligned} \text{Aluminum foil mass (g)} &= \frac{1.00}{4} \text{ g} \\ &= 0.25 \text{ g} \end{aligned}$$

c. 12.5 mL

$$0.25 \text{ g}$$

Chemistry Honors Study Guide

**Norbert J. Pienta, Melanie M.
Cooper, Thomas J. Greenbowe**

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