

Types of Chemical Reactions Worksheet #1

H/Chemistry

For each of the following equations, identify what kind of reaction it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 1) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ **Synthesis**
- 2) $\text{CaI}_2 + \text{Cl}_2 \rightarrow \text{CaCl}_2 + \text{I}_2$ **Single Replacement**
- 3) $3\text{KOH} + \text{AlCl}_3 \rightarrow \text{Al}(\text{OH})_3 + 3\text{KCl}$ **Double Replacement**
- 4) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ **Synthesis**
- *5) $\text{Ca}(\text{OH})_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ **Double Replacement**
- 6) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ **Decomposition**
- 7) $3\text{CuSO}_4 + 2\text{Al} \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{Cu}$ **Single Replacement**
- 8) $\text{Na}_2\text{S} + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{S}$ **Double Replacement**
- 9) $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$ **Synthesis**
- *10) $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ **Single Replacement**

*Remember: "H₂O" can also be thought of as "HOH"

Complete each of the following equations as needed to make it the type of reaction indicated. Be sure to write each formula correctly.

- 11) Double replacement: $\text{Na}_2\text{CrO}_4 + \text{PbCl}_2 \rightarrow 2\text{NaCl} + \text{PbCrO}_4$
- 12) Single replacement: $\text{Cl}_2 + 2\text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$
- 13) Decomposition: $\text{Mg}(\text{ClO}_3)_2 \rightarrow \text{MgCl}_2 + 3\text{O}_2$
- 14) Synthesis: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- 15) Double replacement: $3\text{Ca}(\text{OH})_2 + 2\text{FeCl}_3 \rightarrow 2\text{Fe}(\text{OH})_3 + 3\text{CaCl}_2$
- 16) Single replacement: $\text{Fe} + \text{Cu}(\text{NO}_3)_2 \rightarrow \text{Fe}(\text{NO}_3)_2 + \text{Cu}$ [Assume Fe²⁺]
- 17) Decomposition: $2\text{Hg}_2\text{O} \rightarrow 4\text{Hg} + \text{O}_2$
- 18) Synthesis: $\text{S} + \text{O}_2 \rightarrow \text{SO}_2$
- 19) Double replacement: $\text{AgNO}_3 + \text{KI} \rightarrow \text{AgI} + \text{KNO}_3$
- 20) Single replacement: $\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu}(\text{NO}_3)_2$ [Copper (II) is used here]

Chemistry In Context Answer Key

**Robin Millar, John Leach, Jonathan
Osborne, Mary Ratcliffe**

Chemistry In Context Answer Key:

Chemistry in Context Lucy T. Eubanks,2006 **Chemistry in Context** Conrad L. Stanitski,1999 Introduction to Contextual Maths in Chemistry Fiona Dickinson,Andrew McKinley,2023-01-17 CHEMISTRY STUDENT GUIDES GUIDED BY STUDENTS For any student who has ever struggled with a mathematical understanding of chemistry this book is for you Mathematics is the essential tool for physical scientists We know that confidence in using mathematics early on in a chemistry degree builds a solid foundation for further study However applying the abstract mathematics taught in schools to chemical phenomena is one of the biggest challenges that chemistry students face In this book we take a chemistry first approach We link the mathematics to recognisable chemical concepts building on high school chemistry to facilitate deeper understanding We cover the practical mathematical skills including representation of data as tables and graphs and give an overview of error handling in the physical sciences More advanced mathematical concepts are introduced using calculus to determine kinetic rate laws intermolecular forces and in quantifying energetic change in thermodynamics We also introduce the concept of the complex number and its role in considering quantum wave functions widely used in computational chemistry There are worked examples and problem sets to provide plenty of practise material to build proficiency We also include insights from real students which identify common problem areas and provide the prompts that helped them to overcome these Chemistry Student Guides are written with current students involved at every stage guiding the books towards the most challenging aspects of the topic Chemunity News ,1993 Newsletter for chemistry educators at the elementary high school and college levels The School Science Review ,2003 **Children's Books in Print** R R Bowker Publishing,Bowker,1999-12 **The Reading Context** Dorothy U. Seyler,2000 The Reading Context shows students how reading is a three step process through a clear and easy to remember prepare read respond strategy The book starts where developmental students need to start with an explanation of the context created by author work and reader It also stresses the importance of thinking about who the writer is what the author's purpose is what kind of work students are about to read and what they expect to gain from it Selections have been chosen to hold student interest while representing the kinds of material they will meet in their college assignments textbook material from geology to psychology from advertising to history from health to economics Chapters are filled with short pieces for illustration and practice and conclude with two or three longer selections In each chapter clear explanations of strategies to improve reading are supported by many exercises The text's eleven chapters form three sections Chapters 1-4 comprise the nuts and bolts strategies for reading and introduce the idea of the reading context and active reading In chapters 5-8 students examine an author's use of writing strategies as aids to comprehension and develop their own writing to learn strategies Chapters 9-11 provide students with opportunities to read more widely to study expressive and persuasive writing and explore a variety of works all designed to improve critical thinking skills *Chemistry in Context for Cambridge International AS & A Level* Graham Hill,John Holman,Philippa

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