

Study Guide

Key Concepts

- Concentrations can be measured qualitatively and quantitatively.
- Molarity is the number of moles of solute dissolved per liter of solution.

$$\text{molarity (M)} = \frac{\text{moles of solute}}{\text{liters of solution}}$$

- Molality is the ratio of the number of moles of solute dissolved in 1 kg of solvent.

$$\text{molality (m)} = \frac{\text{moles of solute}}{\text{kg of solvent}}$$



Chemistry Solution Concentration Study Guide Answer Key

N. Sue Barnes



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Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Katherine J. Denniston, 2006 *Chemistry, Student Study Guide* John A. Olmsted, Gregory M. Williams, 2005-02-02

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