

# **Study Guide –Chapter 10**

## **CHM1033**

### **Test # 3: (Chapters 10, 11, and 12)**

1. Indicate whether each of the following statements is characteristic of an acid or a base:
  - a) has a sour taste
  - b) neutralizes bases
  - c) produces  $H^+$  ions in water
  - d) is named potassium hydroxide
  - e) neutralizes acids
  - f) produces  $OH^-$  in water
  - g) has a soapy feel
  - h) turn litmus red
2. Name each of the following acids or bases:
  - a)  $HCl$
  - b)  $Ca(OH)_2$
  - c)  $H_2CO_3$
  - d)  $HNO_3$
  - e)  $H_2SO_3$
  - f)  $Al(OH)_3$
  - g)  $HBr$
  - h)  $H_2SO_4$
  - i)  $KOH$
  - j)  $HNO_2$
3. Write formulas for the following acids and bases.
  - a) magnesium hydroxide
  - b) hydrofluoric acid
  - c) phosphoric acid
  - d) lithium hydroxide
  - e) ammonium hydroxide
  - f) sulfuric acid

# Ch 10 Study Guide For Chemistry

**T. W. Graham Solomons**



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**Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl** Paul B. Kelter, 1993

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