



Alkynes:

C A) Introduction:

1) Structure and physical properties:

-  \Rightarrow no rotation across the $C \equiv C$
 \Rightarrow linear structure
- $C \equiv C$ formed of 1 σ bond and 2 π bond
- $C \equiv C$ triple bond adds 2 unsaturation
 \hookrightarrow formula: acyclic alkynes: C_nH_{2n-2}
so, $CH_3-C \equiv C-CH_3$
- cyclic alkyne: C_nH_{2n-4}

- pKa of terminal alkyne is 25.
 $CH_3-C \equiv C-H \xrightarrow{B^-} CH_3-C \equiv C^- + BH$
terminal alkyne
- Internal alkyne: $CH_3-C \equiv C-CH_3$
- the melting point of alkynes increase with the # of carbons

* common names; terminal alkynes are named as substituted acetylenes

ex:



Systematic: ethyne

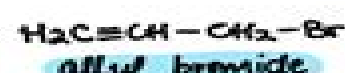
1-butyne

2-pentyne

Common: acetylene

ethyl-acetylene

ethylmethyl-acetylene



* above structures also show comparison between allyl and alkyne *

(2) Nomenclature:

* same rules for alkenes but use suffix "yne" \Rightarrow alkyne

* multiple triple bonds:

2 = diynes; 3 = triynes; 4 = tetra-ynes

ex:



(3E, 9S)-9-chlorodeca-3-ene-1,6-diyne
Stereo-chemical info

- lowest position for alkyne (#1, 6)
- alkene #3
- always mention E/Z, R/S

ex:

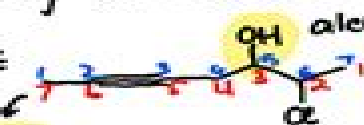


3-chlorocyclooctyne



alcohols and amines have higher priority over alkenes and alkynes.

ex:



name:

2-chlorohept-5-yne-3-ol

alcohol should have lowest priority
not marked with blue but marked with red

ex:



(2S, 3R)-2-chloro-N-methyl-5-heptyn-3-amine

* complex problem (won't be seen on hw/exam)

I doubt

Ch 9 Alkynes Study Guide

Mary H. Bailey



Ch 9 Alkynes Study Guide:

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