

12

STOICHIOMETRY

Practice Problems

In your notebook, solve the following problems.

SECTION 12.1 THE ARITHMETIC OF EQUATIONS

Use the 3-step problem-solving approach you learned in Chapter 1.

11. A car dealer has 100 bags of apples, 20 crates of peaches, and 10 tubs (pounds) of cherries. Write a balanced equation that fits this situation. How many apples are needed to make one peach?
12. Two moles of potassium chloride and three moles of oxygen are produced from the decomposition of two moles of potassium chlorate, $2\text{KClO}_3(s)$. Write the balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate?
13. Using the equation from problem 12, how many moles of oxygen are produced from 1.0 moles of potassium chlorate?
14. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water. How many molecules of water are produced from 2.0×10^{24} molecules of oxygen? How many moles of water are produced from 22.0×10^{24} molecules of oxygen?

SECTION 12.2 CHEMICAL CALCULATIONS

15. Calculate the number of moles of hydrogen chloride produced from 30 moles of hydrochloric acid.



16. Calculate the number of moles of chlorine needed to form 1.0 moles of iron(III) chloride.



17. Calculate the number of grams of nitrogen dioxide that are produced from 4.0 moles of nitric oxide.



18. Calculate the mass, in grams, produced from the decomposition of 75.0 g of potassium chlorate.



19. Calculate the mass, in grams, needed to react with chlorine to produce 164 g of silver chloride. (Refer to Table 1 for the molar mass of each element.)

20. How many liters of carbon monoxide at 500°C are needed to react with 4.00 g of oxygen gas to produce carbon dioxide?



21. Calculate the number of moles of oxygen gas needed to produce 4.00 liters of nitrogen monoxide. Assume all gases are at the same conditions of temperature and pressure.



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Chemistry Antony C. Wilbraham, Dennis D. Staley, Prentice-Hall, Inc, Michael S. Matta, Edward L. Waterman, 2004-08
Resources in Education, 1985 **Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents **Teaching Science for Understanding** James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities *Resources in Education*, 1996 *Prentice Hall Chemistry*, 2000 *Illinois Chemistry Teacher*, 2006 *SourceBook Version 2.1*, 1998

Resources in Vocational Education ,1978

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science in Africa the Arab States Asia and the South Pacific Europe and North America Latin America and the Caribbean and is based largely on the reports and discussions at the meeting The third part contains some examples of topics and modules of integrated science courses taken from recent courses in Botswana the Caribbean the Netherlands the Philippines Sierra Leone and the United Kingdom The fourth part is an annotated bibliography over 370 entries which attempts to sample literature relevant to integrated science KR

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