



Name: _____
Date: _____

COUNTING ATOMS

WORKSHEET

1. Chemical formulas can have three components.

- The _____ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
- The _____ which are numbers that are found on the lower _____ - hand side of each element symbol. This tells you the number of _____ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only _____ atom of this element in the molecule.
- The _____ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be _____ by the _____ on the outside of the bracket found on the lower _____ - hand side of that bracket.

2. Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH

Element	# of Atoms
Total	

2) HNO_3

Element	# of Atoms
Total	

3) U_2O

Element	# of Atoms
Total	

4) U_2SO_4

Element	# of Atoms
Total	

5) NH_4Cl

Element	# of Atoms
Total	

6) CaClO_3

Element	# of Atoms
Total	

7) H_3COCH_3

Element	# of Atoms
Total	

8) $\text{Mg}(\text{OH})_2$

Element	# of Atoms
Total	

9) $\text{Al}(\text{OH})_3$

Element	# of Atoms
Total	

10) $\text{NH}_4\text{C}_2\text{H}_5\text{O}_2$

Element	# of Atoms
Total	

11) $\text{NaC}_2\text{H}_5\text{O}_2$

Element	# of Atoms
Total	

12) $(\text{NH}_4)_2\text{PO}_4$

Element	# of Atoms
Total	

13) $\text{Ca}_3(\text{PO}_4)_2$

Element	# of Atoms
Total	

14) $\text{Fe}_2(\text{CO}_3)_3$

Element	# of Atoms
Total	

15) $\text{Al}_2(\text{SO}_4)_3$

Element	# of Atoms
Total	

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Julie R. Frentrup

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Chemistry, Student Study Guide James E. Brady,Fred Senese,2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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