

DENSITY WORKBOOK**Question:** What is Density?**Answer:** Density is a measure of how much mass is contained in a given unit volume. Put simply, if mass is a measure of how much 'stuff' there is in an object; density is a measure of how tightly that 'stuff' is packed together. The principle of density was discovered by the Greek scientist Archimedes.Overview of Density Worksheet**Key formulas/Concepts:**

- **Density** - Density = Mass divided by Volume ($D = M/V$). The mass of an object is 25 grams. The volume of an object is 5 cm^3 . $D = 25\text{g}/5\text{cm}^3 = 5 \text{ g/cm}^3$.

$$\text{density} = \frac{\text{mass}}{\text{volume}}$$
- **Volume of a rectangular shaped object/cube** - Length X Width X Height ($L \times W \times H$).

Example: A cube has a length of 3 cm. (Note - a cube has six equal sides). $V = 3 \times 3 \times 3 = 27 \text{ cm}^3$. A piece of wood has a length of 10 cm, width of 18 cm, and height of 3 cm. $V = 10 \times 18 \times 3 = 540 \text{ cm}^3$.

- **Volume of an irregularly shaped object** - Use a graduated cylinder filled with a fluid. Record the beginning amount of fluid (in mL - milliliters). Drop the object into the graduated cylinder. Record the level of the fluid with the object (in mL). To determine the volume: **Level of fluid with object (minus) beginning amount of fluid**. Example - a graduated cylinder is filled to the 30mL mark with water. You drop in a rock. The water level rises to 48mL. $V = 48\text{mL} - 30\text{mL} = 18\text{cm}^3$.

KEY NOTE

- ✓ A submerged object displaces a volume of liquid equal to the volume of the object.
- ✓ One milliliter (1 mL) of water has a volume of 1 cubic centimeter (1cm^3).

Problems (Always Show Your Work and units - g, mL, cm^3 , g/cm^3 , g/mL):

1. $M = 34\text{g}$, $V = 17\text{cm}^3$

2. $M = 42\text{g}$, $V = 28\text{cm}^3$

3. You have a box that has a volume of 210 cm^3 and mass of 42g .
 What is its density?

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**John McMurry, Robert A.
Pribush, Robert Fay**

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