

Chemistry Study Guide

Periodic Table of the Elements

1 H Hydrogen	2 He Helium																	18 Ar Argon	19 K Potassium	20 Ca Calcium											36 Kr Krypton	37 Rb Rubidium	38 Sr Strontium											54 Xe Xenon	55 Cs Cesium	56 Ba Barium											86 Rn Radon	87 Fr Francium	88 Ra Radium											118 Og Oganesson
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🎯

Matter:

Anything that has **mass** and **occupies space** ✖️🌐

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** 🧪👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Carbon Chemistry Study Guide

Bernice G. Segal, Peter S. Shenkin



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Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011 *An Introduction to the study of the compounds of carbon, or, Organic chemistry* Ira Remsen, 1907 **Organic Chemistry Study Guide with Solutions Manual** Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999

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S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples **Chemistry, Study Guide** Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry chemical equilibrium and thermochemistry Worked examples now number over 300 and exercises over 1460 **Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition** Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines **Organic Chemistry Study Guide & Problems Book** Brent L. Iverson, 1998 Companion to Organic chemistry 2nd ed by William H Brown Provides a detailed section by section overview of the major points covered in the text All of the problems from the text are reprinted here with detailed stepwise solutions **Study Guide for E. Russell Hardwick's Introduction to Chemistry** Joan Bouillon, 1984 *Study guide to accompany Drew H. Wolfe: General, organic and biological chemistry* Daniel C. Pantaleo, 1986 *Study Guide and Solutions Manual to Accompany The Essence of Organic Chemistry* Jane M. Cram, Donald J. Cram, 1978 **A Reader's Guide to Contemporary Literature** William Swan Sonnenschein, 1896 **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** John McMurry, Susan McMurry, 1986 Study Guide to Organic Chemistry Robert Thornton Morrison, Robert Neilson Boyd, 1992 A popular introduction to organic chemistry which stresses the importance of molecular structure in understanding the properties and principles of organic chemistry Provides a wide variety of spectra to be analyzed Features four color photographs throughout *Solutions Manual and Study Guide to Accompany Introduction to Organic Chemistry, 4th Ed* Paul A. Bartlett, Judith G. Koch, 1992

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