

Chemistry Study Guide

Periodic Table of the Elements																		
H	Li	Be																
Na	Mg																	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
Fr	Ra	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Uut	Fl	Uup	Lv	Uus	Uuo		
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
		Ac	Th	Pa	U	Ng	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

Objectives:

To understand the structure of an atom, and the nature + properties of subatomic particles.



Matter:

Anything that has mass and occupies space.

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the smallest, indivisible particle.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protonz (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles.

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p ⁺	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e ⁻	-0	-1	Orbit/Shell

Carbon Chemistry Study Guide

Bernice G. Segal, Peter S. Shenkin

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Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy,2011 *An Introduction to the study of the compounds of carbon, or, Organic chemistry* Ira Remsen,1907 **Organic Chemistry Study Guide with Solutions**

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S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text. The new edition of the Study Guide includes many new worked out examples. **Chemistry, Study Guide** Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry, chemical equilibrium and thermochemistry. Worked examples now number over 300 and exercises over 1460. **Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition** Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises. Also includes summaries of name reactions, functional group synthesis and reactions, lists of reagents and abbreviations, and articles on topics ranging from infrared absorption frequencies to the Nobel Price winners in Chemistry. This edition now includes all new artwork expanded in text, problems, summary quizzes approximately every three chapters, more detailed explanations in solutions and chapter outlines.

Organic Chemistry Study Guide & Problems Book Brent L. Iverson, 1998 Companion to Organic chemistry 2nd ed by William H Brown. Provides a detailed section by section overview of the major points covered in the text. All of the problems from the text are reprinted here with detailed stepwise solutions. **Study Guide for E. Russell Hardwick's Introduction to Chemistry** Joan Bouillon, 1984 *Study guide to accompany Drew H. Wolfe: General, organic and biological chemistry* Daniel C. Pantaleo, 1986 *Study Guide and Solutions Manual to Accompany The Essence of Organic Chemistry* Jane M. Cram, Donald J. Cram, 1978 **A Reader's Guide to Contemporary Literature** William Swan Sonnenschein, 1896

Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry John McMurry, Susan McMurry, 1986 **Study Guide to Organic Chemistry** Robert Thornton Morrison, Robert Neilson Boyd, 1992 A popular introduction to organic chemistry which stresses the importance of molecular structure in understanding the properties and principles of organic chemistry. Provides a wide variety of spectra to be analyzed. Features four color photographs throughout. *Solutions Manual and Study Guide to Accompany Introduction to Organic Chemistry, 4th Ed* Paul A. Bartlett, Judith G. Koch, 1992

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