

Periodic Trends

1. Explain the following trends:

- Atomic Radius
- Electronegativity
- Ionization Energy

2. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

3. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

4. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

5. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

6. Explain the following trends:

- Electronegativity
- Atomic Radius
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7. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

8. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

9. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

10. Explain the following trends:

- Electronegativity
- Atomic Radius
- Ionization Energy

Name: _____ Date: _____ Period: _____

Periodic Trends Test Study Guide

Section 1: Atomic Radius and Electronegativity

1. Explain the following trends:

- Atomic Radius: Increases from top-left to bottom-right.
- Electronegativity: Increases from bottom-left to top-right.

2. Explain the following trends:

- Atomic Radius: Increases from top-left to bottom-right.
- Electronegativity: Increases from bottom-left to top-right.

3. Explain the following trends:

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10. Explain the following trends:

- Atomic Radius: Increases from top-left to bottom-right.
- Electronegativity: Increases from bottom-left to top-right.

Chemistry Unit Study Guide

Harry Nicholls Holmes



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