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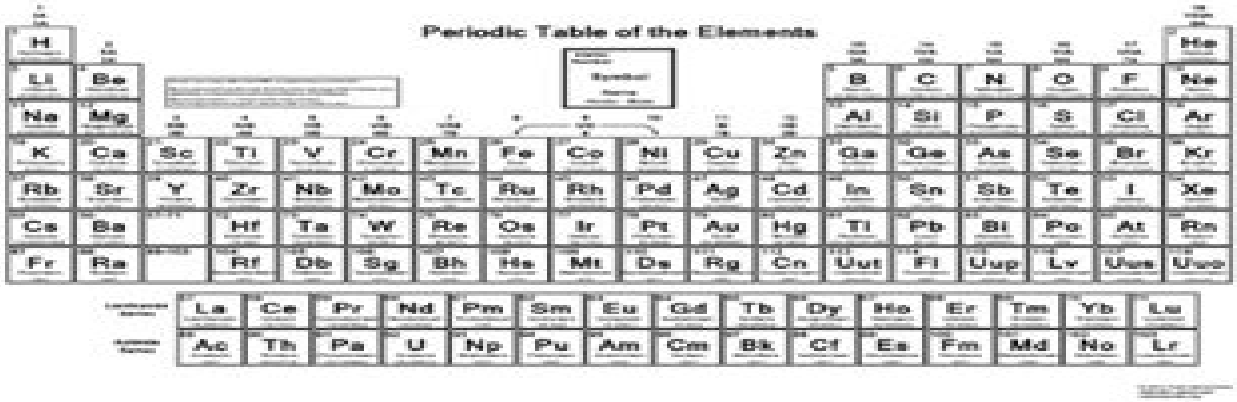

🔬 Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🧑‍🔬✨.

💡 Matter:

Anything that has **mass** and **occupies space** 🏋️🌍

Examples: air, gases, liquids, solidz!

⚛️ Atoms:

Atoms are the **smallest, indivisible particle** 🔬✂️.

- **Outer part** → *shell/orbit/energy level* → carries the electronz (e−)
- **Inner part** → *nucleus* → houses the protons (p+) and neutronz (n)
- Outer & Inner part → together these are the **subatomic particles**

🧪 Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~0	-1	Orbit/Shell

Chemistry Study Guide Review Answers

James E. Brady

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Chemistry, Study Guide James E. Brady,John R. Holum,1993-02-11 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples Fundamentals of Chemistry, Study Guide James E. Brady,John R. Holum,1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs Student Study Guide Peter L. Kresan,Reed Mencke,Frank Press,Raymond Siever,2003-09-25 This reconceptualization of the text Understanding Earth reflects the fundamental changes in the field of physical geology over the past several years *General Chemistry, Study Guide* James E. Brady,1990-03-01 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems Understanding Earth Student Study Guide Peter L. Kresan,Reed Mencke,2006-05-03 The guide helps students prepare for lectures and exams with a heavy emphasis on utlizing the book s Web resources Organic Chemistry, Study Guide T. W. Graham Solomons,1992-01-24 The Fifth Edition has undergone the most extensive revisions of any edition so far Changes include new early chapter on acids and bases in organic chemistry ionic and free radical mechanisms stronger emphasis on the biological environmental medical and industrial applications of organic chemistry as well as on organic synthesis early introduction of the structure and reactivity of carbonyl compounds oxidation reduction reactions and synthesis using Grignard and organolithium reagents **Study Guide with Selected Solutions** Karen Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems General Chemistry Study Guide Sixth Edition Darrell D. Ebbing,1999 *Study Guide for Organic Chemistry* Seyhan N. Eğe,1999 *Basic Concepts of Chemistry, Study Guide* Leo J. Malone,2000 The text s three

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