

Moles and Reactions

Introduction: Molar Ratios in Reactions

Throughout this chapter, we have seen how to balance chemical equations. Now we turn our attention to reactions in which the reactants and products are not in stoichiometric proportions.



For most reactions of propane that we have studied, we have assumed that the reactants are stoichiometric, that is, the amounts of reactants are such that the amounts of products that we expect to form are also stoichiometric. This is not true for many reactions. Specifically, the stoichiometric proportions that we expect to form are not necessarily the amounts of products that will actually form in certain situations and conditions of reaction. These differences are called stoichiometric discrepancies, or stoichiometric imbalances. In a stoichiometric imbalance, the reactants are present in amounts that are not stoichiometric.

The Molar Thinking Checklist

Below, the questions in the molar thinking checklist apply to the following problem:

(a) How many moles of Na_2O_2 are required to completely react with 1.00 mol of H_2O ?



For this reaction, Na_2O_2 and H_2O are present in stoichiometric amounts (1:1 ratio).



(b) How many moles of O_2 are produced when 1.00 mol of Na_2O_2 is added to 1.00 mol of H_2O ?



(c) For the reaction $\text{Na}_2\text{O}_2 + \text{H}_2\text{O} \rightarrow \text{Na}_2\text{OH} + \text{O}_2$, how many moles of O_2 are produced when 1.00 mol of Na_2O_2 reacts with 1.00 mol of H_2O ?



(d) For the reaction $\text{Na}_2\text{O}_2 + \text{H}_2\text{O} \rightarrow \text{Na}_2\text{OH} + \text{O}_2$, how many moles of O_2 are produced when 1.00 mol of Na_2O_2 reacts with 1.00 mol of H_2O ?



(e) How many moles of O_2 are produced in (d)?



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web jan 24 2011 for trumpet in c or bb and piano george enescu 1881 1955 biographical info he was born in the village of liveni later renamed george enescu in his honor dorohoi county at the time today botoșani county he showed musical talent

from early in his childhood a child prodigy enescu created his first musical
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