

NAME _____

Date _____

COUNTING BY MEASURING MASS

Purpose: To determine the mass of several samples of chemical compounds and use the data to count atoms.

Materials: safety goggles, tablespoon measure, weighing paper, small beaker or cup, scale, water, piece of aluminum, table salt (NaCl), baking soda (NaHCO₃).

- Procedure:**
1. Measure the mass of about 10 mL of water.
 2. Measure the mass of the aluminum bar or a large ball of aluminum foil.
 3. Measure the mass of one tablespoon of NaCl.
 4. Measure the mass of one tablespoon of NaHCO₃.
 5. Complete the table below showing all calculations.

Data Table (Show your work on a separate sheet of paper):

	H ₂ O (l)	Al (s)	NaCl (s)	NaHCO ₃ (s)
Mass (grams)				
Molar mass (g/mol)				
Moles of each compound				
Moles of each element				
Atoms of each element				

Counting By Measuring Mass Lab Answer Key

Terry C. Jones

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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