

NAME _____

Date _____

COUNTING BY MEASURING MASS

Purpose: To determine the mass of several samples of chemical compounds and use the data to count atoms.

Materials: safety goggles, tablespoon measure, weighing paper, small beaker or cup, scale, water, piece of aluminum, table salt (NaCl), baking soda (NaHCO₃).

Procedure:

1. Measure the mass of about 10 mL of water.
2. Measure the mass of the aluminum bar or a large ball of aluminum foil.
3. Measure the mass of one tablespoon of NaCl.
4. Measure the mass of one tablespoon of NaHCO₃.
5. Complete the table below showing all calculations.

Data Table (Show your work on a separate sheet of paper):

	H ₂ O (g)	Al (g)	NaCl (g)	NaHCO ₃ (g)
Mass (grams)				
Molar mass (g/mol)				
Moles of each compound				
Moles of each element				
Atoms of each element				

Counting By Measuring Mass Lab Answer Key

Terry C. Jones

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Nuclear Science Abstracts ,1952 Selected Water Resources Abstracts ,1991 **The Essential Lab Manual** Karen Timberlake,2002-06-24 Drawing from the successful main Laboratory Manual the Essential Laboratory Manual includes twenty one experiments which have been revised and updated Suitable for a one or two term lab course *Government Reports Announcements* ,1971-05 **U.S. Government Research and Development Reports** ,1965 **Transactions of the American Nuclear Society** American Nuclear Society,1981 *INIS Atomindex* ,1988 **Physics Briefs** ,1991
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 General, Organic, and Biochemistry Charles H. Henrickson,Charles Henrickson, Ph.D.,2003-03 Government Reports Announcements & Index ,1988 *Lab World* ,1973 Journal of the Textile Institute Textile Institute (Manchester, England),1926 List of members in v 1 8

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