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Dr. Joshua Eckroth



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ec on the basis of this experiment what is the heat of solution per mole of KNO_3 f w 101 11 u the specific heat of water is 4 184 j g k g soln 2 58 g 98 57 g 101 15 g Δt cal 20 4 ec 22 5 ec 2 1 ec

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web in a calorimeter the white solid BaSO_4 forms and the temperature of the mixture increases to 28 1 c assuming that the calorimeter absorbs only a negligible quantity of heat and that the specific heat capacity of the solution is 4 18 j c g and that the density of the final solution is 1 0 g ml calculate the enthalpy change per mole

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