

1. Write the equations depicting the behavior of the following acids and bases in water.

- $\text{HBr} \rightarrow \text{H}^+ + \text{Br}^-$
- $\text{HCN} \rightarrow \text{H}^+ + \text{CN}^-$
- $\text{RbOH} \rightarrow \text{Rb}^+ + \text{OH}^-$
- $\text{KOH} \rightarrow \text{K}^+ + \text{OH}^-$

2. Classify each reactant and product as an acid (A), base (B), conjugate acid (CA), or conjugate base (CB).

- $\text{F}^- + \text{H}_2\text{O} \rightarrow \text{HF} + \text{OH}^-$
- $\text{HClO} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{ClO}^-$
- $\text{H}_3\text{PO}_4 + \text{NH}_3 \rightarrow \text{NH}_4^+ + \text{H}_2\text{PO}_4^-$
- $\text{HNO}_3 + \text{HS}^- \rightarrow \text{H}_2\text{S} + \text{NO}_3^-$

a- base, acid, conjugate acid, conjugate base
b- acid, base, conjugate acid, conjugate base
c- acid, base, conjugate acid, conjugate base
d- acid, base, conjugate acid, conjugate base

3. Classify each of the following acids as monoprotic, diprotic, triprotic, or tetraprotic.

- H_2SO_4 (sulfuric acid) diprotic

- Acetic acid, show to the right



monoprotic

- HCN (cyanic acid) monoprotic

4. Oxaloacetic acid, which is produced in metabolic reactions in the human body has the following structure.



- Classify the above structure as a mono-, di-, tri-, or tetraprotic acid. Explain your answer.

Diprotic acid because it can only transfer 2 hydrogen protons

Ch 9 Practice Problems Chemistry Answers

**Patrick E. McMahon, Rosemary
McMahon, Bohdan Khomtchouk**



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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields.

Chemistry Calculations for Beginners John Obimakinde, Samuel Obimakinde, Ebenezer Obimakinde, Fredrick Akinbolade, 2025-05-30 With decades of combined experience as science teachers at both school and undergraduate levels, the authors have recognised that one of the greatest challenges faced by students studying chemistry is grasping the complexity of the numerous numerical problems found in most parts of the subject. This text is crafted to provide a clear and accessible pathway to overcoming this challenge by assisting students, especially novices or those with minimal knowledge of the subject, in performing chemistry calculations. The content covers fundamental calculations crucial to understanding the principles of chemistry, making it an invaluable tool for students aiming to excel in their studies. Key features: Designed with a student friendly approach including detailed explanation of chemical concepts underlying each type of calculation; step by step explanations; alternative methods for solving problems; numerous practice exercises; answers to practice exercises; and appendices. The book is tailored to suit various curricula, ensuring relevance for a diverse audience. Encompasses a wide range of calculations, offering students a thorough understanding of essential chemistry concepts. Serves as an excellent resource for exam preparation and equips students with skills applicable to future scientific endeavours. Employs straightforward language to ensure ease of understanding for beginners. Uses IUPAC conventions, underscoring the universal nature of chemistry.

General Organic and Biological Chemistry Kenneth W. Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing, dental hygiene, nutrition, medical technology, and occupational therapy. It is also suited for students majoring in other fields where it is important to have an understanding of the basics of

chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features Student's Guide to Chemistry; a Modern Introduction David Brooks,1974 **Student Study Guide Fundamentals of Chemistry by David E. Goldberg** Margaret G. Kimble,1994
Prentice Hall Chemistry ,2000 Problem Solving with Python Michael D. Smith,2026-01-20 An innovative new way to teach computational thinking and problem solving that makes programming accessible to anyone Problem solving with computation has become a basic literacy required of modern life but the traditional way we teach students to code doesn't work for everyone This innovative textbook provides a highly engaging alternative approach Problem Solving with Python is a hands on introduction to computational thinking useful computer science concepts and the art of computer programming where skills and ideas are introduced in service of solving an interesting problem Each chapter begins with an ambiguous problem description drawn from everyday life that resolves with a piece of working code Gradually progressing in difficulty the book's three act structure charts a clear developmental path from novice to skilled programmer Michael Smith first presents the basics of programming through repeated application of a worklist algorithm allowing the reader to become comfortable in problem decomposition and fundamentals before attempting more complicated algorithms and approaches He then shows how to solve real world problems using the power of abstraction algorithms and the right data structures Finally the exercises in the book's last act fully transition the reader from programmer to problem solver Based on the author's popular class at Harvard this accessible textbook builds conceptual understanding through practical skills development to enable anyone to master the what and how of computational thinking Prioritizes the development of computational thinking Does not assume students are intrinsically motivated to learn programming Emphasizes active learning through real world problems and case studies Is suitable for students and self learners from all backgrounds Includes coverage of data representation arithmetic and logical operations algorithms networks computability operating systems and compilers memory systems and security Offers extensive ancillary resources Student Solutions Manual Darrell D. Ebbing,2002-07 Provides solutions to exercises solutions to odd numbered practice problems general problems and cumulative skills problems plus answers to review questions *Understanding Chemistry* Fred M. Dewey,1994 **Preparatory Chemistry** Howard Stephen Stoker,1990 Chemistry John S. Phillips,Cheryl Wistrom,2000 **The American Journal of Science** ,1918 **Chemistry** Steven S. Zumdahl,1997 **Fundamentals of Chemistry** David E. Goldberg,2001 *Holt Chemistry* Ralph Thomas Myers,2004 **Partial Solutions Manual** Darrell D. Ebbing,David Bookin,1998 *Organic Chemistry* Paula Yurkanis Bruice,2001 This book a companion to Bruice's Organic chemistry allows you to introduce modern molecular modeling into your organic chemistry course The heart of the book is a CD ROM containing over 200 models Preface
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